

Building a Physics-Aware AI Ecosystem for Solid-State Hydrogen Storage Materials

Seong-Hoon Jang, Yiwen Yao, Chuanyu Liu, Linda Zhang,* Di Zhang, Xue Jia, Hung Ba Tran, Eric Jianfeng Cheng, Ryuhei Sato, Yusuke Ohashi, Toyoto Sato, Yusuke Hashimoto, Mark D. Allendorf, Nongnuch Artrith, Marcello Baricco, Andreas Borgschulte, Darren P. Broom, Ang Cao, Benjamin Wei Jie Chen, Lixin Chen, Ping Chen, Eun Seon Cho, Stefano Deledda, Zhao Ding, Martin Dornheim, Michael Felderhoff, Yaroslav Filinchuk, George E. Froudakis, Mingxia Gao, Thomas Gennett, Zaiping Guo, Ikutaro Hamada, Jason Hattrick-Simpers, Bjørn C. Hauback, Michael Hirscher, Torben R. Jensen, Baohua Jia, Hyoung Seop Kim, Takahiro Kondo, Kentaro Kutsukake, Xiao-Yan Li, Tongliang Liu, Piao Ma, Jianfeng Mao, Rana Mohtadi, Hyunchul Oh, Mark Paskevicius, Chris J. Pickard, Astrid Pundt, Anibal J. Ramirez-Cuesta, Hiroyuki Saitoh, Kaihang Shi, Aloysius Soon, Chenghua Sun, Chris Wolverton, Hiroshi Yabu, Weijie Yang, Zhenpeng Yao, Xuebin Yu, Jianxin Zou, Shouyi Hu, Panpan Zhou, Xi Lin, Zhigang Hu, Zhenhao Zhou, Pengfei Ou,* Jiayu Peng,* Shin-ichi Orimo,* and Hao Li*

 Cite This: <https://doi.org/10.1021/acseenergylett.6c01856>

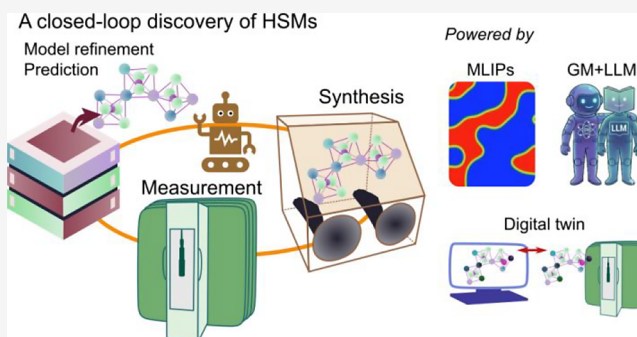
 Read Online

ACCESS |

 Metrics & More

 Article Recommendations

ABSTRACT: Hydrogen storage remains a central bottleneck for scalable hydrogen energy systems due to the multiscale and coupled nature of the thermodynamics, kinetics, and microstructural evolution of hydrogen storage materials (HSMs). Although artificial intelligence (AI) has accelerated materials discovery, current approaches remain constrained by fragmented data, limited physical consistency, and weak integration with experimental validation. Here, we propose a unified framework that integrates coherent data infrastructure, physics-grounded modeling, and AI-driven inverse design within a closed-loop discovery paradigm. By constraining optimization with thermodynamics, kinetics, uncertainty, provenance, and experimental feedback, this approach enables adaptive, physically consistent optimization, thereby establishing a pathway toward autonomous, digital-twin-enabled discovery of HSMs.



A closed-loop discovery of HSMs

Hydrogen is widely regarded as a cornerstone of future carbon-neutral energy systems, yet its deployment remains constrained by the lack of efficient, safe, low-cost, and reversible storage solutions.^{1,2} Its low density under ambient conditions (0.082 kg m^{-3} , corresponding to a volumetric energy density below 10 MJ m^{-3}) necessitates densification strategies for practical use. While compression (35–70 MPa) and liquefaction ($<25 \text{ K}$) are established approaches, both impose significant energy penalties and engineering challenges, limiting scalability. These limitations originate from the intrinsically weak intermolecular interactions of dihydrogen, which hinder dense packing. Here, we focus specifically on solid-state hydrogen storage materials (HSMs), thereby excluding liquid organic hydrogen carriers. Within this framework, two broad strategies have therefore emerged: *physisorption* in high-surface-area

porous solids and *chemisorption* via metal–hydrogen bond formation, each with distinct advantages and limitations.³

Physisorption-based storage relies on weak gas–surface interactions and is the most effective at low temperatures. Nanoporous materials such as metal–organic frameworks (MOFs), covalent–organic frameworks (COFs), hydrogen-bonded organic frameworks (HOFs), and porous organic polymers (POPs) offer high surface areas and tunable pore environments, enabling systematic

Received: June 17, 2026

Revised: July 17, 2026

Accepted: July 30, 2026

exploration of structure–property relationships.^{4–7} However, low adsorption enthalpies limit performance near ambient conditions, motivating efforts to enhance binding strength without sacrificing reversibility.

In contrast, solid-state hydrogen storage via chemisorption in metal and complex hydrides offers significantly higher volumetric densities.^{8–10} In these systems, hydrogen uptake and release involve bond formation and phase transformations, leading to favorable storage capacities but also introducing challenges related to thermodynamics and kinetics. The behavior of such materials is commonly characterized by pressure–composition isotherms, from which key quantities such as enthalpy, entropy, and maximum storage capacity can be extracted. In practice, however, reliable datasets remain scarce due to sluggish kinetics, contamination effects, and sensitivity to microstructural and processing conditions, which hinder both fundamental understanding and the rational design of materials and systems.

Beyond these two archetypes, emerging material classes (including compositionally complex high-entropy alloys) offer additional opportunities to tune hydrogen interactions through multicomponent design.¹¹ Despite this diversity, the discovery of high-performance HSMs remains slow and largely empirical, reflecting the intrinsic complexity of coupled thermodynamic and kinetic processes. In addition, HSMs span distinct chemical classes (i.e., including metal hydrides, complex hydrides, and nanoporous frameworks) that often exhibit limited transferability of design principles, effectively forming “islands” of chemistry. Also, additional classes (i.e., non-reversible chemical hydrides, hydrolysis-based systems, and liquid organic hydrogen carriers) present distinct challenges and design criteria, further broadening the hydrogen storage landscape.

Hydrogen storage performance is governed not only by the capacity but also by the equilibrium pressure, operating temperature, diffusion kinetics, phase transformations, and long-term cycling stability, reflecting a fundamentally multi-objective optimization problem.¹² Crucially, these properties are path-dependent, evolving across multiple length and time scales and remaining highly sensitive to activation history, defect structure, impurities, and operating conditions. In addition, system-level

factors (e.g., heat and mass transport and tank design) impose further constraints on practical performance, collectively limiting the predictive design.¹³

Recent advances in data-driven materials science and artificial intelligence (AI) offer new opportunities to address these challenges.^{14–20} High-throughput computational screening, curated experimental datasets, and machine-learning (ML) models enable rapid exploration of vast design spaces and identification of promising candidates. However, progress is currently constrained by fragmentation across data sources and by the limited integration of physically meaningful descriptors, kinetic effects, and degradation processes. As a result, model robustness and transferability remain limited.

Addressing these challenges requires a more tightly integrated framework that connects data generation, model development, and experimental validation. While automation and self-driving laboratory (SDL) approaches can accelerate data acquisition and improve reproducibility, their primary value lies in enabling the generation of consistent, high-quality datasets that capture condition-dependent behavior. When coupled with physics-informed models, such data can support more reliable predictions and iterative refinement of materials design strategies, rather than serving as an end in themselves.

A new paradigm is therefore emerging that integrates data infrastructure, physics-grounded modeling, and adaptive experimentation within a unified, closed-loop framework. In this context, hydrogen storage remains inherently multiscale and path-dependent, requiring approaches that explicitly account for thermodynamic consistency and reproducibility. The current capabilities, limitations, and open challenges are summarized in Table 1, highlighting key bottlenecks and opportunities for integration across these domains.

Here, based on the current state of the field and in-depth discussions among 69 active researchers from the global communities of HSMs, AI, SDLs, and computational science, we outline a pathway toward an integrated, physics-aware ecosystem for solid-state HSM discovery. In this context, “physics-aware” refers to more than the use of AI models for prediction; it treats thermodynamic consistency, kinetic feasibility, uncertainty,

Table 1. Maturity and Key Challenges of AI-Driven Components in HSM Discovery

component	current capabilities	key limitations	next challenges
Data infrastructure	Curated datasets, high-throughput computation, high-throughput experiments	Fragmentation, datasets with incompatible formats and schemas, inconsistent datapoints across datasets, missing kinetics/degradation, incomplete experimental context, limited reproducibility assessment	Interoperable, multimodal, thermodynamically consistent, and reproducibility-aware datasets with uncertainty, provenance tracking, data augmentation strategies
Physics-grounded modeling with MLIPs	Atomistic dynamics, diffusion, defects, phase transformation	Limited transferability, free-energy accuracy	Integration with thermodynamics, open-system formalisms, and multiscale models under experimentally relevant conditions
GMs	Structural exploration, inverse design	Lack of physical constraints, uncertain synthesizability, limited experimental grounding, out-of-distribution predictions	Physics-constrained generation coupled with thermodynamic validation and synthesis-aware screening
LLMs	Knowledge extraction, hypothesis generation, multimodal literature mining	Lack of physical grounding, hallucinations, alignment, loss of context, variable reliability, limited treatment of uncertainty	Integration with physical models and databases, provenance-aware reasoning frameworks
Active learning and uncertainty quantification	Active learning for computations, Bayesian optimization for experimentation	Separate acquisition frameworks for experiments and computations, different data fidelities	Hybrid, cost-aware experimental–computational acquisition strategies
Closed-loop discovery	ML-guided experiments, early automation	Limited adaptability, integration gaps, weak uncertainty handling, insufficient data-quality awareness	Autonomous, adaptive experimental systems, linked to reproducibility-aware databases and confidence-calibrated decision-making

provenance, and experimental feedback as integral constraints throughout data construction, modeling, inverse design, and validation. This distinguishes it from conventional materials informatics workflows, where physical validation is often applied only after candidate prediction. By combining structured data, interpretable models, and iterative experimental validation, this approach aims to accelerate the identification and optimization of HSMs, ultimately enabling scalable and efficient energy systems.

The central bottleneck in data-driven hydrogen storage research is no longer the lack of predictive algorithms, but the absence of a thermodynamically coherent and interoperable data infrastructure grounded in experiment. Hydrogen storage data exist in two partially disconnected layers (Figure 1a): (i) computational repositories containing density functional theory (DFT) energies, structures, and descriptors and (ii) heterogeneous experimental reports, where key information is often buried in figures or narrative text, spread across formats and languages, and is difficult to retrieve with standard search methods. This fragmentation forces reliance on expert interpretation and limits reliable ML and inverse design.

Computational repositories have advanced significantly. Databases such as the Materials Project, OQMD, AFLOW, NOMAD, and Alexandria provide large-scale access to computed structures and energetics, enabling high-throughput screening.^{21–25} However, these datasets are biased toward equilibrium properties under idealized conditions and generally lack surface and interface information. For solid-state HSMs, where performance depends on pressure-temperature behavior, kinetics, and cycling stability, static computational data provide an incomplete representation. Moreover, the lack of explicit linkage between computed entries and experimentally validated systems limits their applicability.

For nanoporous materials such as MOFs, high-throughput simulations have enabled large databases linking structure, pore characteristics, and adsorption behavior (e.g., CoRE, QMOF, and MOFX-DB).^{26–28} These datasets more directly connect structure to performance under defined conditions, although still largely within idealized frameworks. Despite this progress, structure–property relationships remain limited by kinetic effects, phase transformations, and cycling complexity that are not fully captured in such models. More broadly, experimental and theoretical studies of MOFs and COFs have established key principles governing hydrogen adsorption.^{4,29} While pristine frameworks can achieve high gravimetric uptake under cryogenic conditions, their performance near ambient conditions is constrained by weak host–guest interactions. Efforts to enhance adsorption enthalpy (through alkali metal doping, transition metal functionalization, and electronic structure tuning) have demonstrated improved binding, but achieving practical storage performance under realistic conditions remains challenging.³⁰ Recent work combining high-throughput screening and ML is therefore shifting the focus toward rational design strategies that target both adsorption energetics and operating conditions.³¹ Recent developments further incorporate data-driven approaches for synthesizability prediction and transfer learning across adsorption spaces in MOFs, enabling more efficient exploration of experimentally accessible materials.^{32,33}

Curated experimental databases have improved accessibility and consistency, partially bridging the gap between computational repositories and real systems. For example, the Digital Hydrogen Platform (*DigHyd*: www.dighyd.org) extracts and

standardizes hydrogen storage data from the literature (Figure 1b), aggregating thousands of publications and entries focused on thermodynamic quantities derived from pressure–composition–temperature measurements.³⁴ In *DigHyd*, each standardized entry is linked to the source publication, material identity, measurement type, extracted or reported value, unit convention, reference state, and data provenance, enabling comparison across heterogeneous literature records. Such efforts address a key limitation: capacity alone is insufficient to evaluate performance, as thermodynamic and kinetic constraints ultimately determine usability. However, inconsistencies in reported quantities (e.g., hydrogen content definitions and reference states) continue to introduce uncertainty and complicate comparison. In addition, both computational and experimental datasets typically report single-point values without quantified uncertainties.

AI-assisted data extraction has enabled the construction of such datasets at scale. Frameworks such as the descriptive interpretation of visual expression (DIVE) approach enable systematic extraction of figure-based data, including pressure–composition isotherms and desorption profiles.³⁵ However, automated extraction introduces challenges related to accuracy, hallucination, and reproducibility, particularly when relying on large language models (LLMs). Retrieval-augmented approaches can partially mitigate these issues, but robust validation remains essential.

A more fundamental limitation lies in the absence of standardized data reporting protocols. Experimental hydrogen storage data are frequently presented in graphical form, resulting in the loss of underlying measurement information and limiting reproducibility. This issue is particularly acute for microporous materials, where key adsorption features occur at very low relative pressures and are not easily extracted. Establishing machine-readable formats, analogous to the adsorption information file (AIF),³⁶ alongside standardized reporting guidelines for hydrogen sorption measurements,³⁷ would significantly improve data comparability and reuse. Furthermore, many hydrogen storage reactions involve amorphous or partially disordered intermediates that are not readily captured by periodic representations, highlighting the need for approaches that incorporate local structural descriptors or statistical models of disorder.

Future hydrogen-storage databases should therefore record not only extracted values but also minimum contextual metadata, including PCT curves, temperature and pressure protocols, activation procedures, catalyst identity and loading, gas purity, kinetic conditions, cycling history, uncertainty, and provenance. Such information is essential because nominally similar HSMs can show different behaviors depending on the processing history, impurities, catalysts, and measurement conditions.

Scalability alone does not resolve the core challenge: hydrogen storage data are highly sensitive to experimental context, including the activation history, processing history, microstructure, impurities, catalysts, surface and mechanical stress, and cycling conditions. Importantly, impurities and compositional inhomogeneities can lead to the formation of secondary phases, which fundamentally alter the thermodynamic behavior rather than acting as minor perturbations. As a result, datasets lacking detailed contextual metadata can propagate errors into derived parameters and ML models.³⁸ Addressing this requires a shift toward reproducibility-aware data curation, where each record

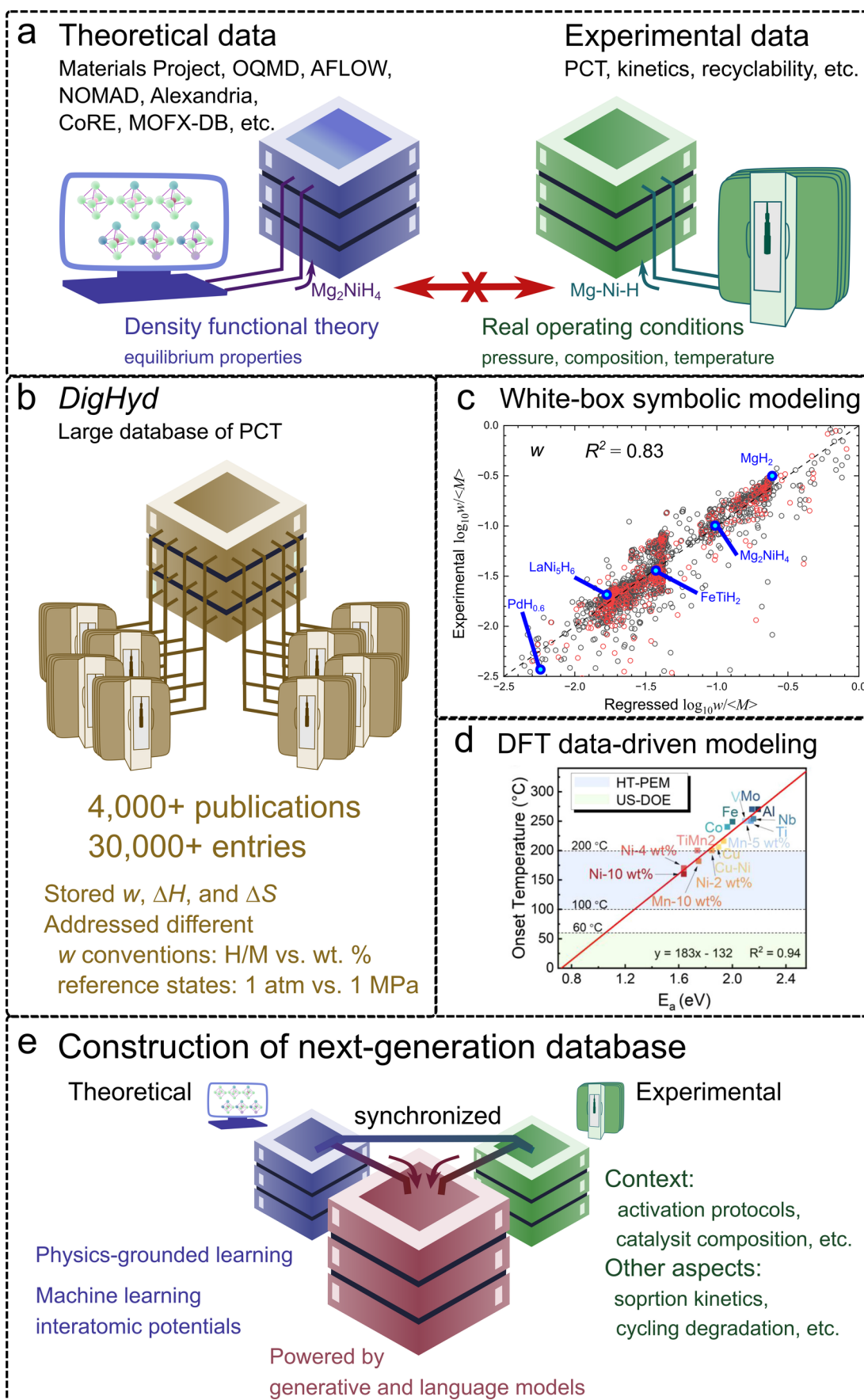


Figure 1. continued

Figure 1. Data infrastructure for solid-state HSMs. (a) Schematic comparison between computational and experimental data. Computational databases (e.g., Materials Project, OQMD, AFLOW, NOMAD, Alexandria, CoRE, QMOF, and MOFX-DB)^{21–28} provide DFT-derived equilibrium properties under idealized conditions, whereas experimental data reflect real operating environments, including pressure, temperature, compositional variability, impurities, surface conditions, measurement uncertainty, and bulk morphology. The lack of direct correspondence between these domains highlights a fundamental gap in the current data infrastructure. (b) Representative example of curated experimental data from the Digital Hydrogen Platform (*DigHyd*), comprising >4000 publications and >30,000 entries of thermodynamic measurements (e.g., hydrogen capacity w , enthalpy ΔH , and entropy ΔS).³⁴ Standardization across differing reporting conventions (e.g., H/M vs wt % and reference states) was addressed. (c) *DigHyd* enables the application of white-box symbolic modeling (as an example, a symbolic modeling case of gravimetric storage density (w) is represented), revealing physically interpretable relationships between composition and hydrogen storage properties.^{39–41} Reproduced from ref 39, under the terms of CC BY-NC 4.0. (d) Data-driven regression linking dehydrogenation barrier (E) and onset temperature for MgH₂-based materials, highlighting the gap between the current material performance and US Department of Energy (US-DOE) and the high-temperature proton-exchange membrane (HT-PEM) targets.⁴² Reproduced from ref 42, under the terms of CC BY-NC 4.0. (e) Conceptual framework for a next-generation hydrogen storage database integrating computational and experimental data. The unified platform incorporates thermodynamic, kinetic, degradation-related and reproducibility-aware metadata, together with experimental context such as activation protocols, catalyst composition, gas purity, cycling history, and measurement provenance. Explicit treatment of uncertainty, negative/null results, and confidence-weighted data quality enables more reliable comparison between literature records and computational predictions. Physics-grounded learning and MLIPs enhance predictive consistency, while GMs and LLMs facilitate data integration, knowledge extraction, and inverse design.

includes experimental provenance, measurement conditions, and confidence levels.

Such a framework should also distinguish between directly reported and extracted data, incorporate uncertainty quantification, and apply physics-based consistency checks across thermodynamic quantities. Importantly, negative and null results should be retained, as they define failure boundaries and improve model calibration. In future AI-driven HSM platforms, uncertainty and reproducibility should be treated as central design principles that determine which data are trusted, which models are deployed, and which experiments should be performed next. This progression transforms static databases from passive repositories into digital data platforms that actively support model updating, confidence scoring, and closed-loop candidate selection.

Despite progress, the current data landscape remains fragmented. While computational datasets provide structural descriptors and curated platforms organize thermodynamic data, other critical properties (e.g., kinetics, cycling degradation, hysteresis, activation barriers, impurity sensitivity, and phase evolution) remain sparse and inconsistently reported. This imbalance limits model robustness and transferability.

This fragmentation directly constrains predictive modeling. ML models trained on incomplete datasets may perform well within limited domains but fail to generalize across materials and conditions. In contrast, curated datasets with physically meaningful descriptors enable more reliable and interpretable models. For example, symbolic regression applied to curated hydrogen storage datasets has revealed compact, interpretable relationships between composition and thermodynamic properties (Figure 1c).^{39–41} Similarly, data-driven analyses of Mg-based hydrides quantify the dehydrogenation behavior and highlight gaps relative to application targets such as those defined by the US Department of Energy (Figure 1d).^{42,43}

A critical next step is the integration of experimental and computational data into a unified framework (Figure 1e). On the experimental side, standardized identifiers, reproducibility-aware metadata, consistent reporting of key properties such as microstructure and phase constitution, and inclusion of negative results are required. Consistent reporting of microstructure and phase constitution are particularly important as microstructural features strongly influence kinetics. On the computational side,

advances in physics-based learning and machine-learning interatomic potentials (MLIPs) offer improved representation of realistic behavior across time and length scales. Together, these developments point toward a hydrogen-storage-specific data ecosystem that is interoperable, physically grounded, and predictive.

HSMs are often evaluated in terms of equilibrium capacity and hydride formation stability, yet their practical performance is governed by dynamics.⁴⁴ Hydrogen diffuses through solids, redistributes among interstitial sites, interacts with defects, and drives the nucleation and evolution of hydride phases during cycling.⁴⁵ These processes are further influenced by volume changes and internal stresses, which can enhance transport or induce hysteresis, fracture, and degradation. Hydrogen storage is therefore governed by evolving atomic and microstructural states rather than static structures. Notably, hydrogen-containing systems often exhibit significant dynamical disorder, with some phases stabilized only at finite temperature, as reflected by imaginary phonon modes in static calculations; capturing such effects, including their impact on ionic mobility, remains a key challenge for both data representation and atomistic modeling.

This distinction has direct implications for atomistic modeling. DFT has been indispensable for evaluating hydrogen binding, migration barriers, and defect energetics, but is limited to small-scale, largely static simulations. Ab initio molecular dynamics (AIMD) extends this framework but remains restricted to short trajectories and cannot capture converged diffusion, phase boundaries, or long-time degradation. Mesoscale approaches describe phase-front propagation, chemo-mechanical coupling, and hysteresis, but depend on atomistic inputs that are often inconsistent and valid only over limited regimes.⁴⁶ A central challenge is therefore achieving DFT-level accuracy at experimentally relevant spatial and temporal scales.

MLIPs provide a critical bridge (Figure 2a).⁴⁷ By learning potential-energy surfaces from DFT data, MLIPs enable simulation of hydrogen motion, structural rearrangement, and mechanical response across extended scales. They replace explicit electronic-structure calculations with surrogate models that predict energies, forces, and stresses from atomic environments.⁴⁸ This capability is particularly important for solid-

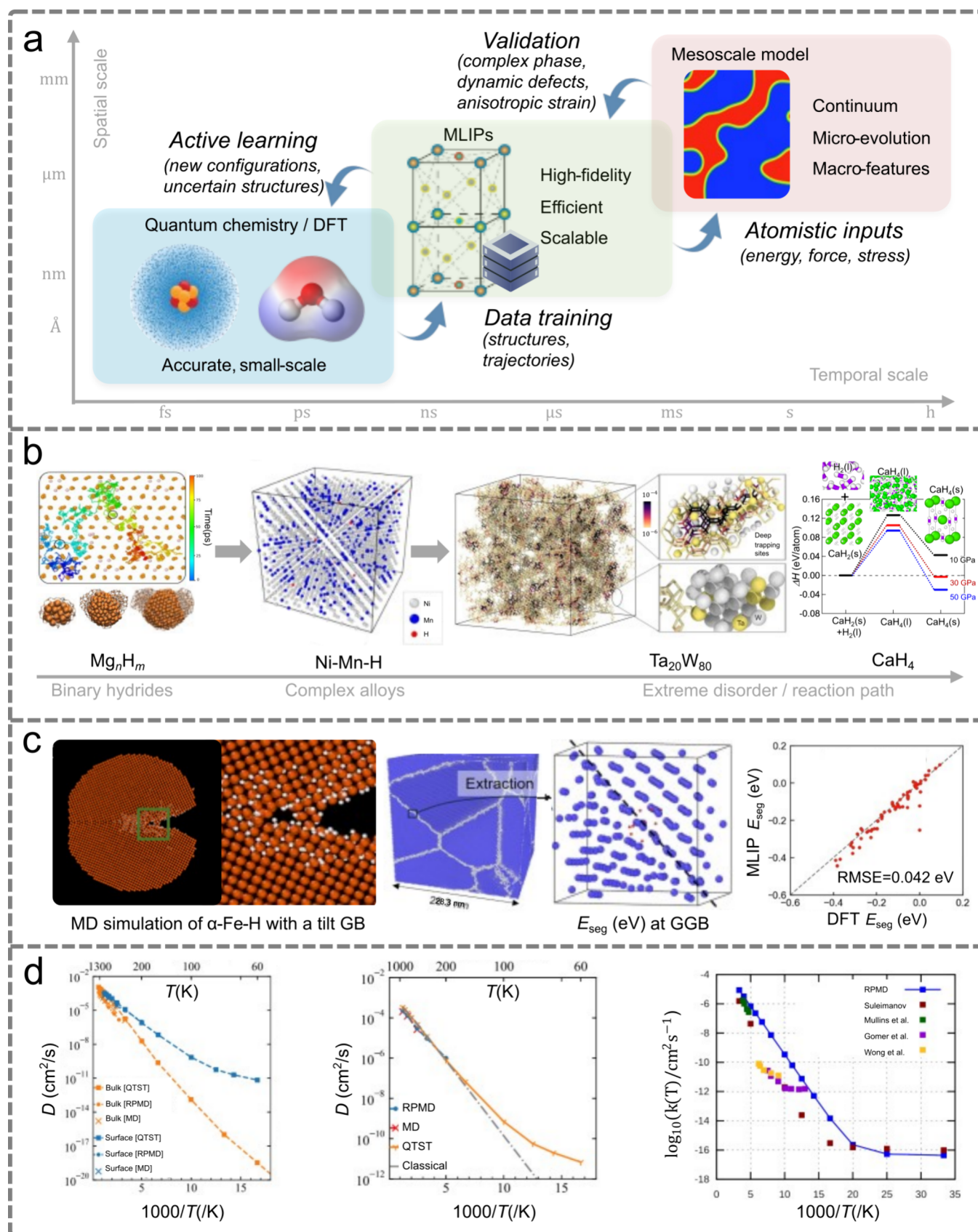


Figure 2. MLIP-enabled multiscale modeling for HSMs. (a) Multiscale modeling. MLIPs serve as links bridging the spatiotemporal gap. Trained on quantum-fidelity but expensive data, MLIPs achieve DFT accuracy at the speed of classical MD simulations and continuously self-refine via active learning with labeled uncertain configurations. MLIPs feed atomistic data into mesoscale models, which provide validation of complex phases,

Figure 2. continued

dynamic defects, and anisotropic strain, before supplying quantum-informed parameters into macroscopic applications. (b) Modeling of hydrogen diffusion across environments. *Left*: Time-progression trajectory (blue to red) of a hydrogen atom in $\text{MgH}_{0.0625}$ at 673 K, illustrating oscillations within interstitial sites.⁵² Reproduced from ref 52 under the terms of CC BY 4.0. Equilibrated Mg–H nanoclusters at 300 K, where orange spheres and bicolor sticks denote unbonded Mg and Mg–H coordination, respectively.⁵³ Reproduced with permission from ref 53. Copyright 2020 American Physical Society. *Center*: Representative atomic configuration of a Ni–25 at.% Mn alloy used in MD to predict hydrogen diffusion coefficients.⁵⁴ Reproduced from ref 54 under the terms of CC BY 4.0. *Right*: Atomistic mechanisms of super-Arrhenius diffusion in non-equimolar $\text{Ta}_{20}\text{W}_{80}$ at 500 K; the visualization maps H visitation frequencies at tetrahedral interstitial sites, highlighting the local metallic environments of specific trapping and inaccessible sites.⁵⁵ In addition, MLIP-driven MD simulations capture surface-melting-driven hydrogen absorption pathways during high-pressure hydrogenation for CaH, providing direct insights into the reaction mechanisms beyond equilibrium diffusion.⁵⁹ Reproduced from refs 55 and 59, under the CC BY 4.0. (c) Atomistic insights into hydrogen segregation and microvoid evolution at grain boundaries (GBs). *Left*: Visualization of microvoid nucleation (green box) within an α -Fe system containing a tilt GB at a 1.856% hydrogen concentration.⁶⁰ Reproduced from ref 60, under CC BY 4.0. *Right*: Representative calculation cell showing the initial positions of H atoms (red) relative to Fe atoms (blue) used to evaluate segregation energies (E_{seg}) at general grain boundaries (GGB). The E_{seg} values for 85 distinct sites at GGB show good agreement between MLIP (vertical axis) and DFT (horizontal).⁶¹ Reproduced from ref 61, under CC BY 4.0. (d) Temperature (T) dependence and quantum effects of hydrogen diffusion on the Pd(111) surface and bulk. *Left*: Pd(111) surface diffusion is faster than bulk Pd at low T . This disparity arises from environment-specific nuclear quantum effects and competing zero-point energy (ZPE) corrections in the surface and bulk. Data are evaluated using the quantum transition state theory (QTST), ring polymer molecular dynamics (RPMD), and classical MD.⁶⁴ *Center*: Surface diffusion D on the Pd(111) evaluated using QTST, RPMD, classical MD, and classical transition state theory. The classical method exhibits Arrhenius behavior (a straight line). In contrast, the path-integral simulations (QTST, RPMD) deviate from this straight line below 200 K, showing a clear non-Arrhenius behavior.⁶⁴ Reproduced from ref 64, under CC BY 4.0. *Right*: An MLIP model coupled with RPMD and kinetic Monte Carlo (kMC) resolved T -dependent hydrogen diffusion coefficients on Ni(100) and literature comparison.⁶⁵ Reproduced with permission from ref 65. Copyright 2020 American Institute of Physics.

state HSMs, where wide composition ranges, competing phases, and defect-rich environments require models that remain reliable across diverse configurations.⁴⁹ Active learning has become central to MLIP development, iteratively expanding the training domain through targeted DFT calculations.⁵⁰

Foundation models have further expanded this capability.⁵¹ These models aim for transferability across chemistries, reducing the need for system-specific training. However, hydrogen storage systems (characterized by defects, multiphase behavior, and anisotropy) remain challenging and often require system-specific validation.

A central application is hydrogen diffusion, a primary kinetic bottleneck. In magnesium hydrides, MLIP-based simulations extend diffusion modeling beyond static barriers and short AIMD trajectories.⁵² These approaches reveal nanoscale transport and structural evolution inaccessible to conventional methods (Figure 2b, left).⁵³ In high-entropy alloys, MLIPs capture environment-dependent diffusion behavior arising from chemical disorder (Figure 2b, center).⁵⁴ Coupled with neural-network-driven kMC, they further reveal super-Arrhenius diffusion in systems such as MoNbTaW (Figure 2b, right).⁵⁵

Beyond transport, MLIPs enable modeling of phase transformations and hydrogen ordering.⁵⁶ In TiCr_2 Laves phases, actively learned MLIPs combined with DFT and Monte Carlo methods predict hydrogen ordering and thermodynamics.⁵⁷ Multiphase MLIPs spanning HCP, BCC, and FCC titanium hydrides demonstrate that accurate diffusion modeling requires broad phase coverage.⁵⁸ MLIP-driven MD has further revealed complex hydrogenation pathways, including surface-melting-driven absorption and high-pressure superhydride formation (Figure 2b, right).⁵⁹

Realistic hydrogen storage behavior also depends critically on defects, surfaces, and interfaces. Defects trap hydrogen, surfaces might hinder and interfaces redirect transport, and cyclic lattice expansion generates stresses that influence kinetics and durability. MLIP-based simulations capture these coupled effects, including hydrogen embrittlement and grain-boundary behavior (Figure 2c).^{60,61} These mechanisms are central to long-term

cycling performance,⁶² particularly in heterogeneous systems such as reactive hydride composites and catalyzed complex hydrides, where interfaces further increase the dimensionality of the design space. Complementary experimental validation is therefore essential: neutron scattering techniques, with their unique sensitivity to hydrogen, enable direct observation of site occupancy, diffusion, and phase evolution, providing key benchmarks for atomistic simulations.

Hydrogen further challenges classical descriptions due to nuclear quantum effects. Zero-point motion and tunneling influence diffusion barriers and Arrhenius prefactors, as well as isotope dependence (Figure 2d). MLIPs enable tractable path-integral simulations,⁶³ capturing phenomena such as non-Arrhenius quantum diffusion on Pd(111) and H diffusion on Ni surfaces.^{64,65}

Despite these advances, current MLIPs remain limited in scope and thermodynamic consistency. Important failure modes include extrapolation outside the training domain, incomplete treatment of long-range interactions, magnetic-state changes, and inaccurate free-energy differences in hydrogen-rich or multiphase systems. Their accuracy often degrades as the hydrogen concentration changes, defects accumulate, or multiphase transformations occur—conditions central to real operation.⁵⁸ Therefore, MLIP predictions for HSMs should be monitored by uncertainty estimates and validated against thermodynamic, kinetic, and experimental observables. Hydrogen storage is governed by free-energy landscapes arising from coupled electronic, structural, and magnetic contributions.⁶⁶ Capturing these landscapes, rather than local forces alone, is essential for predicting experimentally observable behavior.

Progress therefore requires integrating MLIPs with explicit thermodynamic formalisms, including thermodynamic integration, free-energy methods, grand canonical ensembles, and CALPHAD-based approaches to predict plateau pressures, phase coexistence, and hysteresis.⁶⁷ Multiscale coupling provides a complementary pathway.⁶⁸ MLIP-derived quantities (e.g., diffusion coefficients, energetics, and interfacial

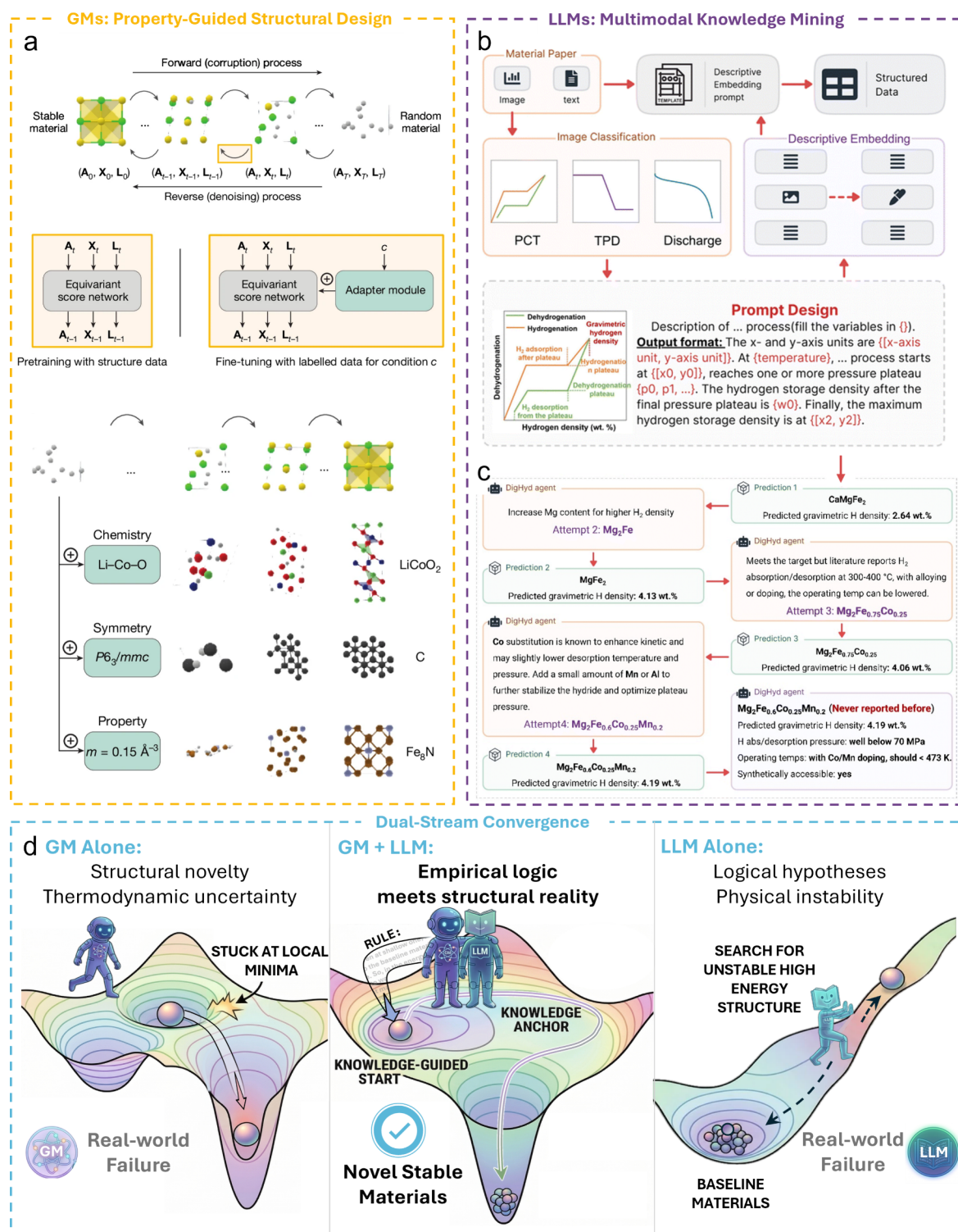


Figure 3. The dual-stream GM–LLM architecture and agent-driven iterative inverse design for the constrained discovery of high-performance HSMs. (a) GMs: property-guided structural design. GMs (e.g., MatterGen) navigate the continuous latent space to propose novel atomic configurations conditioned on user-defined metrics, such as specific chemistry, crystal symmetry, and targeted physical properties; in the context of hydrogen storage, this framework could be extended to incorporate constraints such as the ΔH window.⁷² Reproduced from ref 72 under the terms of CC BY 4.0. (b) LLMs: multimodal knowledge mining. The descriptive interpretation of visual expression workflow processes unstructured scientific literature, translating complex multimodal data into structured databases containing empirical synthesis constraints and thermodynamic relations.³⁵ Reproduced from ref 35, under the terms of CC BY-NC 4.0. (c) LLM agents: iterative inverse design. Building upon the mined historical database, the DigHyd agent executes an autonomous design-prediction-optimization loop.³⁵ The agent proposes initial candidates, which are dynamically evaluated by a pretrained ML regressor. The agent then iteratively reasons and refines the composition (e.g., evolving from CaMgFe_2 to a novel $\text{Mg}_2\text{Fe}_{0.6}\text{Co}_{0.25}\text{Mn}_{0.2}$ alloy) within minutes until researcher-defined thermodynamic targets are met. Reproduced from ref 35, under the terms of

Figure 3. continued

CC BY-NC 4.0. (d) Conceptual illustration of dual-stream convergence between GMs and LLMs for HSM discovery. The integrated GM+LLM framework combines structural exploration with knowledge-based constraints, enabling knowledge-guided navigation of the energy landscape toward physically viable and synthetically accessible stable materials.

properties) can parameterize mesoscale models, enabling simulation of nucleation, coarsening, and phase evolution.⁶⁹ Increasingly, these properties are recognized as size- and state-dependent, evolving with local structure and composition.^{70,71}

Importantly, most current case studies rely heavily on theoretical calculations. While these provide critical mechanistic insights, they remain limited by approximations in electronic structure methods and by incomplete representation of experimental complexity. A key future direction is therefore the integration of active learning with high-quality experimental characterization (e.g., atomic-scale structural analysis and in situ hydrogenation/dehydrogenation measurements) to iteratively calibrate simulations and refine underlying physical models. Such an integration offers a pathway not only to improving the predictive accuracy but also to reconstructing theoretical understanding based on the experimentally validated behavior.

The next generation of multiscale frameworks will therefore combine DFT-level accuracy, MLIP-enabled atomistic dynamics, and mesoscale modeling into a unified, physics-grounded hierarchy capable of capturing microstructural evolution and long-term degradation under realistic conditions.

Beyond predictive fidelity, design-oriented modeling also benefits from explainability. For solid-state HSMs, it is often insufficient to reproduce capacity or diffusion alone; models must relate predictions to chemically meaningful local environments to guide experimental design. Structure-aware approaches provide one route, enabling site-resolved interpretation of model predictions within extended frameworks.

MLIPs have transformed hydrogen storage modeling from static descriptions of binding and barriers to dynamic descriptions of transport, phase transformation, and mechanics. The next challenge is their integration into a thermodynamically coherent, multiscale framework that links atomic-scale processes to macroscopic performance, while enabling the inverse design of solid-state HSMs.

The discovery of high-performance HSMs can be framed as a constrained inverse design problem. Candidate materials must satisfy multiple competing criteria, including capacity, equilibrium pressure, kinetics, and cycling stability, with the requirement of falling within a narrow thermodynamic window representing a key—but not exclusive—constraint for reversible operation under practical conditions. Importantly, both enthalpy and entropy contribute to this constraint; in metal hydrides, these quantities are often coupled, such that reducing thermodynamic stability does not necessarily translate into improved operating conditions. Traditional high-throughput screening struggles to meet these requirements, as it is restricted to existing computational and experimental datasets.⁴⁴

To address this limitation, recent approaches explore a “dual-stream” AI framework combining generative models (GMs) and LLMs. GMs act as structural generators, proposing candidate materials in a continuous configuration space, while LLMs extract empirical knowledge and constraints from unstructured literature.

By learning structural patterns of known materials, GMs generate candidate structures from latent representations, expanding exploration beyond enumerated databases. This capability has enabled the identification of previously unreported structures, including new Mg–H phases and property-conditioned crystal generation using diffusion-based models such as MatterGen (Figure 3a).⁷² However, because these models remain constrained by their training distributions and representations, they often explore variations of known structural motifs. More critically, while recent approaches incorporate property conditioning or post hoc filtering, the enforcement of physical validity, synthesizability, and hydrogen-storage-specific constraints during generation remains limited, which can lead to metastable or non-viable candidates, particularly in large combinatorial spaces such as nanoporous materials.⁷³

In parallel, LLMs are transforming how materials knowledge is utilized. Automated workflows such as DIVE convert graphical data (e.g., pressure-composition-temperature curves) into structured datasets, enabling large-scale database construction (Figure 3b).³⁵ Beyond data extraction, LLMs are increasingly integrated into active learning workflows, where they assist in proposing and refining candidate materials (either in combination with surrogate models or as direct decision-making agents), thereby accelerating iterative materials discovery (Figure 3c).^{74,75} However, LLM-generated hypotheses are not inherently constrained by physical laws and require external validation.

In the near term, LLMs are most realistic as tools for literature mining, data extraction, provenance tracking, synthesis-condition summarization, and human-in-the-loop hypothesis generation, whereas GMs can expand candidate diversity before physics-based validation. Fully autonomous LLM/GM agents that propose, validate, synthesize, and refine HSMs remain a longer-term vision requiring stronger physical constraints and closed-loop experimental validation.

These complementary limitations motivate integration. In a unified framework, LLMs provide empirically grounded constraints (e.g., composition limits, element selection, symmetry preferences, synthesis conditions, hydrogen occupancy ranges, and target property windows), while GMs generate explicit structural candidates that satisfy these inputs (Figure 3d). Coupling this framework with physical validators (e.g., DFT- or MLIP-based simulations) enables thermodynamically consistent screening before experimental validation.^{76,77} Ultimately, embedding such architectures within closed-loop workflows linking design, simulation, and experiment offers a pathway toward autonomous, physics-aware discovery of solid-state HSMs.

The discovery of HSMs is constrained by the interplay between high-dimensional compositional spaces and experimental variables, particularly in compositionally complex systems such as high-entropy alloys, where vast combinatorial spaces and disorder render exhaustive exploration infeasible. Unlike conventional equilibrium materials design, hydrogen storage performance depends not only on the composition

and structure but also on synthesis pathways, microstructural evolution, reaction mechanisms, and cycling behavior. As a result, models trained on static datasets often fail to capture this dynamic, path-dependent behavior, motivating closed-loop discovery frameworks that integrate AI with automated experimentation.

To enable such integration, experimental validation must match the throughput and consistency of computational screening, especially when exploring chemical spaces exceeding 10^6 combinations. SDLs address this need by integrating robotic synthesis, automated characterization, and adaptive monitoring into unified workflows, enabling continuous data generation and standardized measurement.⁷⁸ Integrated platforms for HSMs are emerging, including efforts at Tohoku University under the Green Technologies of Excellence (GteX) program toward autonomous experimental workflows.

Within these frameworks, prediction, synthesis, characterization, and model refinement are iteratively linked (Figure 4a). ML models propose candidates, automated platforms synthesize materials, and standardized experiments (e.g., pressure–composition–temperature measurements and *operando* characterization) provide feedback. Although such data are routinely collected, they are often incompletely reported, limiting reuse. These data refine models, and similar closed-loop approaches have accelerated discovery in other domains, including catalysis and inorganic synthesis.⁷⁹ However, current systems remain largely tool-centric, with limited adaptability to uncertainty or evolving conditions.

The next stage therefore requires a shift from automation to intelligent planning (Figure 4b). Rather than selecting candidates solely based on predicted performance, closed-loop systems must incorporate uncertainty and prioritize experiments

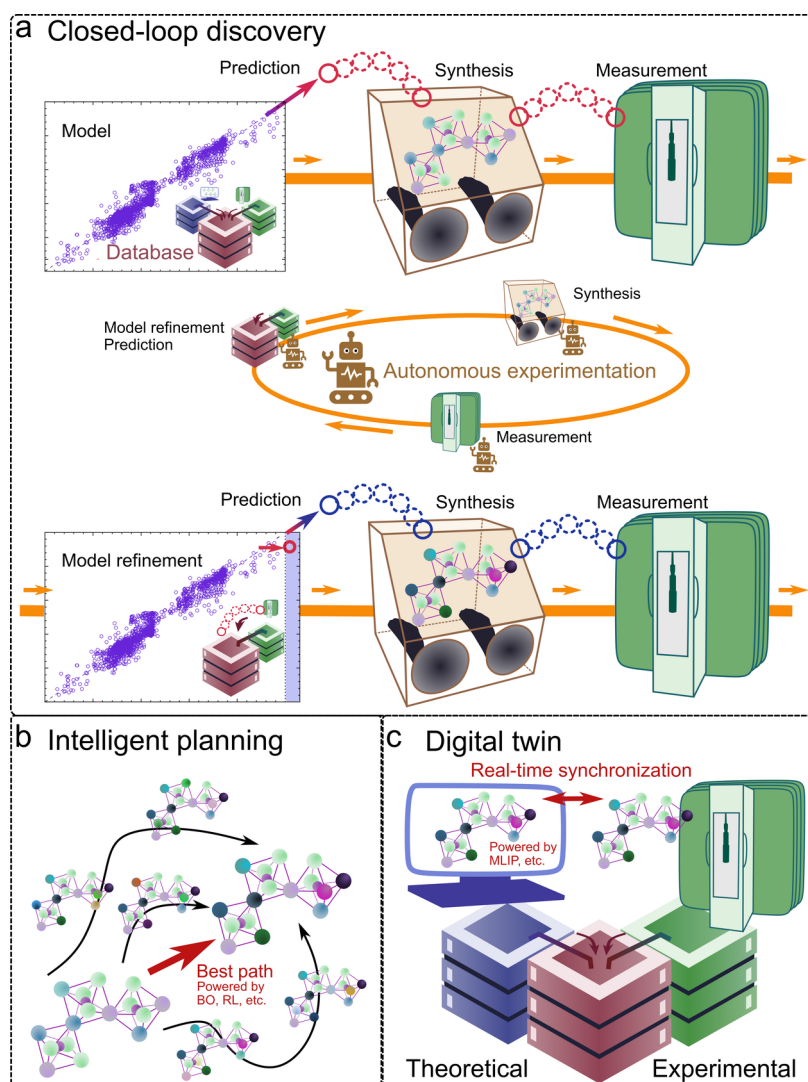


Figure 4. Closed-loop discovery framework for HSMs. (a) Iterative closed-loop workflow for autonomous experimentation, integrating ML-driven prediction, materials synthesis, and experimental measurement. Data generated from each iteration are fed back into the model, enabling continuous refinement and improved predictive accuracy. (b) Intelligent planning in high-dimensional compositional space. Adaptive strategies, such as Bayesian optimization (BO) and reinforcement learning (RL), guide the selection of candidate materials by balancing exploration and exploitation, thereby identifying optimal pathways toward improved hydrogen storage performance. (c) Concept of a digital twin for hydrogen storage systems. A dynamically updated virtual representation, powered by MLIPs and related models, synchronizes computational predictions with experimental observations, enabling real-time feedback, model refinement, and physically consistent interpretation across theoretical and experimental domains.

that improve model reliability. Bayesian optimization enables identification of regions where additional data most effectively reduce uncertainty,^{79,80} while reinforcement learning supports adaptive experiment selection by balancing exploration and exploitation.⁸¹ This is particularly relevant for hydrogen storage, where datasets are uneven: properties such as capacity are relatively well characterized, whereas kinetics, cycling stability, and impurity sensitivity remain sparse. Effective frameworks must therefore allocate experimental effort toward reducing uncertainty in physically meaningful quantities, such as equilibrium pressure and degradation pathways.

Another critical challenge is the integration of heterogeneous data. Hydrogen storage experiments generate diverse outputs, including pressure–composition–temperature isotherms, kinetic measurements, structural characterization, and *operando* spectroscopy, each with different levels of abstraction and uncertainty. Closed-loop systems must therefore incorporate multimodal data fusion, enabling consistent interpretation across data types. This requirement reinforces the need for standardized data representations and robust data infrastructure.

A key direction is the development of a hydrogen storage “digital twin”, a virtual representation that is dynamically updated and synchronizes computational predictions with real-time experimental observations (Figure 4c). At present, such approaches are most naturally applicable within a given class of materials, and extending them across chemically distinct systems remains an open challenge. By integrating MLIPs, thermodynamic models, and kinetic descriptions with continuously updated data, such systems enable the co-evolution of simulation and experiment. Discrepancies between prediction and observation necessitate model refinement, while real-time feedback enables adaptive experimental control and early detection of degradation mechanisms.

Realizing this vision requires addressing several challenges. Experimental data must be sufficiently standardized and reproducible for real-time integration, while MLIPs and surrogate models must accurately capture the thermodynamic and kinetic behavior across diverse systems. In addition, computational and experimental components must be tightly coupled through robust interfaces to enable seamless feedback. These requirements highlight that closed-loop discovery is not solely an algorithmic problem but a system-level challenge spanning data, models, and experimental platforms.

Despite these challenges, closed-loop discovery represents a shift from static, dataset-driven research to dynamic, feedback-driven optimization. For hydrogen storage, where performance is highly sensitive to experimental conditions and governed by coupled thermodynamic and kinetic processes, this approach provides a pathway toward efficient and physically informed exploration. Ultimately, success depends on integrating three key elements: reliable data infrastructure, physics-grounded models, and adaptive decision-making. When combined within a unified framework, these elements enable a transition toward systematic and autonomous optimization of HSMs.

This Perspective highlights the fact that the discovery of high-performance HSMs is fundamentally a system-level challenge that cannot be addressed through isolated advances in data, modeling, or experimentation. Progress is constrained less by the absence of algorithms than by fragmented data, limited physical fidelity across scales, insufficient attention to reproducibility, and weak integration between prediction and

experiment. Addressing these challenges requires a shift toward a cohesive and interoperable research framework.

A key requirement is a thermodynamically consistent data infrastructure. Despite advances in computational and experimental databases, incomplete integration and inconsistent reporting continue to limit predictive reliability. Standardized ontologies, reporting protocols, and explicit links between datasets are therefore essential. At the modeling level, physics-grounded approaches (particularly MLIPs) bridge electronic structure accuracy and large-scale dynamics but must be coupled with thermodynamic formalisms and multi-scale frameworks. In parallel, integrating GMs and LLMs offers new opportunities for inverse design, provided predictions remain physically validated. Ultimately, closed-loop discovery frameworks that integrate data, models, and automated experimentation provide a pathway toward autonomous, physically consistent discovery of next-generation solid-state HSMs. Over the next decade, such physics-aware and digital-twin-enabled workflows could shift solid-state hydrogen storage from experience-driven trial and error toward a reproducible, adaptive, and continuously learning discovery science.

■ AUTHOR INFORMATION

Corresponding Authors

Linda Zhang – Advanced Institute for Materials Research (WPI-AIMR), Tohoku University, Sendai 980-8577, Japan; Frontier Research Institute for Interdisciplinary Sciences (FRIS), Tohoku University, Sendai 980-0845, Japan; Email: linda.zhang.a3@tohoku.ac.jp

Pengfei Ou – Department of Chemistry, National University of Singapore, Singapore 117549, Singapore; orcid.org/0000-0002-3630-0385; Email: pengf.ou@nus.edu.sg

Jiayu Peng – Department of Materials Design and Innovation, University at Buffalo, Buffalo, New York 14260, United States; Email: jypeng@buffalo.edu

Shin-ichi Orimo – Advanced Institute for Materials Research (WPI-AIMR), Tohoku University, Sendai 980-8577, Japan; Institute for Materials Research, Tohoku University, Sendai 980-8577, Japan; Email: shin-ichi.orimo.a6@tohoku.ac.jp

Hao Li – Advanced Institute for Materials Research (WPI-AIMR), Tohoku University, Sendai 980-8577, Japan; orcid.org/0000-0002-7577-1366; Email: li.hao.b8@tohoku.ac.jp

Authors

Seong-Hoon Jang – Advanced Institute for Materials Research (WPI-AIMR), Tohoku University, Sendai 980-8577, Japan; Unprecedented-Scale Data Analytics Center, Tohoku University, Sendai 980-8578, Japan; orcid.org/0000-0001-6026-636X

Yiwen Yao – Department of Chemistry, National University of Singapore, Singapore 117549, Singapore; orcid.org/0009-0007-1780-2789

Chuanyu Liu – Department of Materials Design and Innovation, University at Buffalo, Buffalo, New York 14260, United States

Di Zhang – Advanced Institute for Materials Research (WPI-AIMR), Tohoku University, Sendai 980-8577, Japan; Frontier Research Institute for Interdisciplinary Sciences (FRIS), Tohoku University, Sendai 980-0845, Japan; orcid.org/0000-0001-6347-9344

- Xue Jia** – Advanced Institute for Materials Research (WPI-AIMR), Tohoku University, Sendai 980-8577, Japan
- Hung Ba Tran** – Advanced Institute for Materials Research (WPI-AIMR), Tohoku University, Sendai 980-8577, Japan
- Eric Jianfeng Cheng** – Advanced Institute for Materials Research (WPI-AIMR), Tohoku University, Sendai 980-8577, Japan; orcid.org/0000-0003-0904-802X
- Ryuhei Sato** – Department of Materials Engineering, The University of Tokyo, Tokyo 113-8656, Japan; orcid.org/0000-0002-5503-0982
- Yusuke Ohashi** – Institute for Materials Research, Tohoku University, Sendai 980-8577, Japan
- Toyoto Sato** – Institute for Materials Research, Tohoku University, Sendai 980-8577, Japan; orcid.org/0000-0002-0527-1235
- Yusuke Hashimoto** – Advanced Institute for Materials Research (WPI-AIMR), Tohoku University, Sendai 980-8577, Japan; Frontier Research Institute for Interdisciplinary Sciences (FRIS), Tohoku University, Sendai 980-0845, Japan
- Mark D. Allendorf** – Department of Chemistry, Washington University in St. Louis, St. Louis, Missouri 63130, United States; orcid.org/0000-0001-5645-8246
- Nongnuch Artrith** – Debye Institute for Nanomaterials Science, Utrecht University, 3584 CC Utrecht, The Netherlands; orcid.org/0000-0003-1153-6583
- Marcello Baricco** – Department of Chemistry and NIS, University of Turin, 10125 Turin, Italy; orcid.org/0000-0002-2856-9894
- Andreas Borgschulte** – Advanced Institute for Materials Research (WPI-AIMR), Tohoku University, Sendai 980-8577, Japan; Empa Swiss Federal Laboratories for Materials Science and Technology, Chemical Energy Carriers and Vehicle Systems Laboratory, 8600 Dübendorf, Switzerland; orcid.org/0000-0001-6250-4667
- Darren P. Broom** – Hiden Isochema Ltd, Warrington WA5 7TS, U.K.; orcid.org/0000-0002-1328-7376
- Ang Cao** – State Key Laboratory of Clean Energy Utilization, College of Energy Engineering, Zhejiang University, Hangzhou 310027, China; orcid.org/0000-0003-3787-6949
- Benjamin Wei Jie Chen** – Institute of High Performance Computing, Agency for Science, Technology and Research, Singapore 138632, Singapore; orcid.org/0000-0003-3695-727X
- Lixin Chen** – School of Materials Science and Engineering, Zhejiang University, Hangzhou 310027, China; orcid.org/0000-0002-9624-918X
- Ping Chen** – Dalian Institute of Chemical Physics, Chinese Academy of Sciences, Dalian 116023, China; orcid.org/0000-0002-0625-0639
- Eun Seon Cho** – Department of Chemical and Biomolecular Engineering, Korea Advanced Institute of Science and Technology (KAIST), Daejeon 34141, Republic of Korea; orcid.org/0000-0001-9428-1269
- Stefano Deledda** – Department for Hydrogen Technology, Institute for Energy Technology (IFE), 2007 Kjeller, Norway; orcid.org/0000-0002-4694-1012
- Zhao Ding** – National Engineering Research Center for Magnesium Alloys, College of Materials Science and Engineering, Chongqing University, Chongqing 400044, China; orcid.org/0000-0003-1794-0392
- Martin Dornheim** – Advanced Materials Research Group, Faculty of Engineering, University of Nottingham, Nottingham NG7 2RD, U.K.
- Michael Felderhoff** – Max-Planck-Institut für Kohlenforschung, 45470 Mülheim an der Ruhr, Germany
- Yaroslav Filinchuk** – Institute of Condensed Matter and Nanosciences (IMCN), UCLouvain, 1348 Louvain-la-Neuve, Belgium; orcid.org/0000-0002-6146-3696
- George E. Froudakis** – Department of Chemistry, University of Crete, Voutes Campus, GR-70013 Heraklion, Greece; orcid.org/0000-0002-6907-1822
- Mingxia Gao** – State Key Laboratory of Silicon and Advanced Semiconductor Materials, School of Materials Science and Engineering, Zhejiang University, Hangzhou, Zhejiang 310058, China; orcid.org/0000-0002-4719-6747
- Thomas Gennett** – Colorado School of Mines, Golden, Colorado 80401, United States; orcid.org/0000-0003-1259-6588
- Zaiping Guo** – Department of Materials Science and Engineering, City University of Hong Kong, Kowloon, Hong Kong, SAR 999077, China; orcid.org/0000-0003-3464-5301
- Ikutaro Hamada** – Department of Precision Engineering, Graduate School of Engineering, The University of Osaka, 2-1 Yamadaoka, Suita, Osaka 565-0871, Japan; Research Center for Precision Engineering, Graduate School of Engineering, The University of Osaka, 2-1 Yamadaoka, Suita, Osaka 565-0871, Japan; orcid.org/0000-0001-5112-2452
- Jason Hattrick-Simpers** – Department of Materials Science & Engineering, University of Toronto/CanmetMATERIALS, Toronto, ON M5S 3E4, Canada; orcid.org/0000-0003-2937-3188
- Bjørn C. Hauback** – Department for Hydrogen Technology, Institute for Energy Technology (IFE), 2007 Kjeller, Norway; orcid.org/0000-0002-2717-6941
- Michael Hirscher** – Advanced Institute for Materials Research (WPI-AIMR), Tohoku University, Sendai 980-8577, Japan; Max Planck Institute for Solid State Research, 70569 Stuttgart, Germany; orcid.org/0000-0002-3143-2119
- Torben R. Jensen** – iNANO & Department of Chemistry, Aarhus University, DK-8000 Aarhus, Denmark; orcid.org/0000-0002-4278-3221
- Baohua Jia** – Centre for Atomaterials and Nanomanufacturing, RMIT University, Melbourne, VIC 3001, Australia; orcid.org/0000-0002-6703-477X
- Hyoung Seop Kim** – Graduate Institute of Ferrous Technology (GIFT), Pohang University of Science and Technology (POSTECH), Pohang 37673, Republic of Korea; orcid.org/0000-0002-3155-583X
- Takahiro Kondo** – Advanced Institute for Materials Research (WPI-AIMR), Tohoku University, Sendai 980-8577, Japan; Department of Materials Science, Institute of Pure and Applied Sciences, University of Tsukuba, Tsukuba 305-8573, Japan; Hydrogen Boride Research Center, Tsukuba Institute of Advanced Research, University of Tsukuba, Tsukuba 305-8577, Japan; orcid.org/0000-0001-8457-9387
- Kentaro Kutsukake** – Institute of Materials and Systems for Sustainability (IMaSS), Nagoya University, Nagoya 464-8603, Japan; orcid.org/0000-0001-7878-347X
- Xiao-Yan Li** – Department of Chemistry, National University of Singapore, Singapore 117549, Singapore

Tongliang Liu – School of Computer Science, Faculty of Engineering, University of Sydney, Sydney, NSW 2006, Australia

Piao Ma – Suzhou MatSource Technology Co. Ltd., Suzhou, Jiangsu 215000, China; Gusu Laboratory of Materials, Suzhou, Jiangsu 215000, China

Jianfeng Mao – School of Chemical Engineering, University of Adelaide, Adelaide, SA 5005, Australia; orcid.org/0000-0002-4787-4261

Rana Mohtadi – Toyota Research Institute of North America (TRI-NA), Ann Arbor, Michigan 48105, United States

Hyunchul Oh – Department of Chemistry, Ulsan National Institute of Science and Technology (UNIST), Ulsan 44919, Republic of Korea; orcid.org/0000-0001-6258-0002

Mark Paskevicius – Department of Physics and Astronomy, Institute for Energy Transitions, Curtin University, Perth, WA 6845, Australia; orcid.org/0000-0003-2677-3434

Chris J. Pickard – Advanced Institute for Materials Research (WPI-AIMR), Tohoku University, Sendai 980-8577, Japan; Department of Materials Science & Metallurgy, University of Cambridge, Cambridge CB3 0FS, U.K.

Astrid Pundt – Institute of Applied Materials (IAM-WK), Karlsruhe Institute of Technology (KIT), 76131 Karlsruhe, Germany

Anibal J. Ramirez-Cuesta – Neutron Technologies Division, Oak Ridge National Laboratory (ORNL), Oak Ridge, Tennessee 37831, United States; orcid.org/0000-0003-1231-0068

Hiroyuki Saitoh – Kansai Institutes for Photon Science, National Institutes for Quantum Science and Technology (QST), Hyogo 679-5148, Japan

Kaihang Shi – Department of Chemical and Biological Engineering, University at Buffalo, The State University of New York, Buffalo, New York 14260, United States; orcid.org/0000-0002-0297-1746

Aloysius Soon – Department of Materials Science and Engineering, Yonsei University, Seoul 03722, Republic of Korea; orcid.org/0000-0002-6273-9324

Chenghua Sun – Computational Materials and Catalysis Group, School of Science, Swinburne University of Technology, Melbourne, VIC 3122, Australia; orcid.org/0000-0001-7654-669X

Chris Wolverton – Department of Materials Science and Engineering, Northwestern University, Evanston, Illinois 60208, United States; orcid.org/0000-0003-2248-474X

Hiroshi Yabu – Advanced Institute for Materials Research (WPI-AIMR), Tohoku University, Sendai 980-8577, Japan; orcid.org/0000-0002-1943-6790

WeiJie Yang – Department of Power Engineering, North China Electric Power University, Baoding, Hebei 071003, China; orcid.org/0000-0002-0232-1129

Zhenpeng Yao – Shanghai Key Laboratory of Hydrogen Science & Center of Hydrogen Science, School of Materials Science and Engineering, Shanghai Jiao Tong University, Shanghai 200240, China; Innovation Center for Future Materials, Zhangjiang Institute for Advanced Study, Shanghai Jiao Tong University, Shanghai 201203, China; orcid.org/0000-0001-8286-8257

Xuebin Yu – Department of Materials Science, Fudan University, Shanghai 200433, China; orcid.org/0000-0002-4035-0991

Jianxin Zou – Shanghai Key Laboratory of Hydrogen Science & Center of Hydrogen Science, School of Materials Science and Engineering, Shanghai Jiao Tong University, Shanghai 200240, China; orcid.org/0000-0003-1041-121X

Shouyi Hu – Shanghai Key Laboratory of Hydrogen Science & Center of Hydrogen Science, School of Materials Science and Engineering, Shanghai Jiao Tong University, Shanghai 200240, China

Panpan Zhou – State Key Laboratory of Silicon and Advanced Semiconductor Materials, School of Materials Science and Engineering, Zhejiang University, Hangzhou, Zhejiang 310058, China; orcid.org/0000-0003-0592-5443

Xi Lin – Shanghai Key Laboratory of Hydrogen Science & Center of Hydrogen Science, School of Materials Science and Engineering, Shanghai Jiao Tong University, Shanghai 200240, China

Zhigang Hu – Shanghai Key Laboratory of Hydrogen Science & Center of Hydrogen Science, School of Materials Science and Engineering, Shanghai Jiao Tong University, Shanghai 200240, China; Innovation Center for Future Materials, Zhangjiang Institute for Advanced Study, Shanghai Jiao Tong University, Shanghai 201203, China; orcid.org/0000-0003-1916-6484

Zhenhao Zhou – Shanghai Key Laboratory of Hydrogen Science & Center of Hydrogen Science, School of Materials Science and Engineering, Shanghai Jiao Tong University, Shanghai 200240, China; Innovation Center for Future Materials, Zhangjiang Institute for Advanced Study, Shanghai Jiao Tong University, Shanghai 201203, China

Complete contact information is available at:

<https://pubs.acs.org/doi/10.1021/acsenergylett.6c01856>

Author Contributions

H.L. conceived the overall framework of the Perspective and initiated the international collaboration. S.-H.J. led the manuscript development, coordinated the multi-author drafting and revision process, integrated extensive coauthor input, reconciled overlapping and sometimes divergent revisions, reorganized the manuscript structure, and synthesized the final narrative into a unified Perspective. S.-H.J., Y.Y., C.L., P.O., and J.P. prepared the initial manuscript draft. S.-H.J., P.O., J.P., S.-i.O., and H.L. supervised the project. The manuscript was subsequently developed through an iterative, cross-manuscript collaborative revision process rather than through strictly separated section-by-section contributions. The broader author team provided substantive intellectual input across the manuscript, including critical revision of the scientific framing, correction, and refinement of technical arguments, evaluation of the roadmap, and validation of the Perspective from complementary areas of expertise in hydrogen storage thermodynamics, hydride chemistry, nanoporous materials, experimental reproducibility, pressure–composition–temperature measurements, kinetics and cycling stability, neutron and *operando* characterization, computational materials science, MLIPs, GMs, LLMs, SDLs, and closed-loop materials discovery. Each author critically revised the manuscript for important intellectual content within their area of expertise and contributed to the final scientific interpretation and positioning of the Perspective. All authors approved the final version and agree to be accountable for the

contributions attributed to them. M.B., S.-i.O., and H.L. contributed to funding acquisition.

Notes

The authors declare no competing financial interest.

Biographies

Seong-Hoon Jang is an Associate Professor at Tohoku University. His research focuses on applied mathematics for physics and chemistry, including nonlinear symbolic regression for complex systems, high-throughput MD and structure optimization, and large-Hilbert-space quantum perturbation theory to uncover interpretable mathematical laws governing complex physical and chemical phenomena.

Yiwen Yao is a PhD student in the Department of Chemistry at the National University of Singapore. She obtained her B.E. from Sun Yat-sen University and M.S. in Materials Science from NUS. Her research focuses on combining ML and high-throughput screening to discover high-entropy materials for electrocatalysis.

Chuanyu Liu is pursuing a PhD at SUNY Buffalo. He earned his BS from Chongqing University and MS from the University of Chicago, followed by a full Research Associate position at MIT Chemistry. His research leverages quantum chemistry and ML to develop science-centric AI solutions for materials.

Linda Zhang is an Assistant Professor at the Frontier Research Institute for Interdisciplinary Sciences (FRIS), Tohoku University, Japan. She received her PhD from the Max Planck Institute in 2020. Her research focuses on renewable energy, particularly the development of novel porous materials for hydrogen energy, gas separation, and energy storage.

Di Zhang is a Distinguished Assistant Professor at The Frontier Research Institute for Interdisciplinary Sciences (FRIS), Tohoku University, Japan. His work centers on digital catalysis platforms, catalytic databases, ML, and AI agents for data-driven materials discovery.

Xue Jia is an Assistant Professor at WPI-AIMR, Tohoku University, Japan. She received her PhD from Harbin Institute of Technology (Shenzhen) in 2022. Her research interests focus on AI-driven materials discovery, particularly ML, LLMs, and AI agents for the design and discovery of energy materials.

Hung Ba Tran is an Assistant Professor at the Advanced Institute for Materials Research, Tohoku University. His research focuses on first-principles calculations, Monte Carlo simulations, and ML-assisted materials discovery for magnetic materials and HSMs.

Eric Jianfeng Cheng, Associate Professor at Tohoku University, earned dual PhDs in Materials Engineering and Energy Chemistry. His research centers on solid-state batteries and AI for materials; he pioneered direct visualization of Li dendrite propagation along garnet electrolyte grain boundaries.

Ryuhei Sato is an Assistant Professor at the Department of Materials Engineering at the University of Tokyo. His research interest is mainly in theoretical modeling for inorganic synthesis.

Yusuke Ohashi is an Assistant Professor at the Institute for Materials Research, Tohoku University. He received his PhD in materials engineering from Tohoku University. His research focuses on HSMs, metal hydrides, and microstructure control in functional metallic materials.

Toyoto Sato is an Associate Professor at the Institute for Materials Research, Tohoku University, in Japan. He earned his PhD in

structural chemistry in 2006 from Stockholm University in Sweden. His research interests include the development of HSMs and characterization of their properties and crystal structures.

Yusuke Hashimoto is a Specially Appointed Associate Professor at the Frontier Research Institute for Interdisciplinary Sciences (FRIS), Tohoku University. Holding a PhD in Science, he develops data-driven materials science methods that integrate experimental and computational data through graph-based ML to accelerate the discovery of energy and functional materials.

Mark Allendorf is an Honorary Senior Visiting Researcher in the Chemistry Department at Washington University in St. Louis. He is a former Co-Director of the Hydrogen Materials—Advanced Research Consortium (HyMARC) at Sandia National Laboratories. His research focuses on hydrogen storage and the science and applications of MOFs.

Nongnuch Artrith is an Assistant Professor at Utrecht University. She received her PhD from RUB Germany and held research positions at MIT, UC Berkeley, and Columbia University before joining Utrecht. Her research develops AI and atomistic simulations, integrating computational predictions with experimental data to accelerate the discovery and design of sustainable energy materials.

Marcello Baricco obtained a PhD in Chemistry in 1987. He is a Full Professor in Materials Science and Technology at the Department of Chemistry, University of Torino. He has been coordinator in several research projects with research institutions and industrial partners. He is an expert in Task 51—IEA-TCP Hydrogen.

Andreas Borgschulte is a physicist with focus on hydrogen in materials; PhD in physics from TU Braunschweig (D) in 2002; and Postdoc VU Amsterdam (NL) and Hereon (D); since 2006, group leader of the group Hydrogen Spectroscopy at Empa (CH); lecturer at UZH and ETHZ (CH).

Darren P. Broom is a Product Manager for Hiden Isochema. He received his PhD from the University of Salford, UK, in 2003. His research interests focus on accurately characterizing the hydrogen storage and gas sorption properties of materials. He is currently an Associate Editor of *Adsorption*, *Journal of the International Adsorption Society*.

Ang Cao is a Professor at Zhejiang University. Her research focuses on AI-accelerated catalyst design, multiscale simulations of heterogeneous catalysis, flexible synthesis of green fuels, and experimental studies of ammonia synthesis from nitrogen hydrogenation.

Benjamin W. J. Chen is a materials scientist at A*STAR's Institute of High Performance Computing. He is passionate about using automation, high-throughput simulations, and ML to bridge the gap between computational predictions and experimental validation, so as to discover new materials for securing our energy future.

Lixin Chen is a Full Professor at Zhejiang University, China. He also serves as the deputy director of the Hydrogen Energy Professional Committee of China Renewable Energy Society. His research interests include metal hydrides, multifunctional hydride devices, hydrogen isotope separation and storage systems, novel energy materials for secondary batteries, etc.

Ping Chen received her PhD from Xiamen University. She was a faculty member of the National University of Singapore and is now a Professor at the Dalian Institute of Chemical Physics, Chinese Academy of Sciences. She and her research team are dedicated to exploring hydride materials for hydrogen storage, ammonia synthesis, and hydride ion conduction.

Eun Seon Cho is an Associate Professor of Chemical and Biomolecular Engineering at KAIST. She received her PhD from MIT in 2013 and completed postdoctoral research at LBNL. Her research focuses on functional hybrid nanomaterials for energy and environmental applications, especially hydrogen energy.

Stefano Deledda is Principal Scientist at the Institute for Energy Technology (IFE), Norway. His research focuses on metal hydrides for hydrogen storage, advanced energy materials, and hydrogen technologies. He has extensive experience in metallic materials research and currently coordinates two European research projects on hydrogen storage in metal hydrides.

Zhao Ding is an Associate Professor in the College of Materials Science and Engineering at Chongqing University and an Associate Editor of the Journal of Energy Storage. His research focuses on magnesium-based solid-state hydrogen storage, metal hydrides, and interfacial and catalytic engineering for reversible hydrogen storage.

Martin Dornheim is Leverhulme International Professor for Hydrogen Storage Materials and Systems and Task Leader of the IEA Hydrogen TCP Task 51: Hydrogen Materials for Energy Storage. His research focuses on HSMs, metal hydrides, and hydrogen technologies, advancing sustainable energy through innovative materials design and energy storage systems.

Michael Felderhoff is a research group leader at the Max Planck Institut für Kohlenforschung in Mülheim an der Ruhr, Germany. His research topic is on material development for energy and hydrogen storage, with focus on magnesium compounds, complex hydrides, and high-entropy alloys.

Yaroslav Filinchuk is a Professor of Chemistry at the Université catholique de Louvain (UCLouvain), Belgium. His research focuses on structural chemistry, crystallography, porous materials, metal hydrides, and HSMs, employing advanced diffraction techniques to elucidate structure–property relationships for sustainable energy applications.

George E. Froudakis is a Professor of Physical Chemistry at the University of Crete. His research focuses on computational chemistry, porous materials, hydrogen storage, nanomaterials, and AI-driven materials discovery, developing multiscale modeling and ML approaches to design advanced materials for energy and environmental applications.

Mingxia Gao is a Professor in the School of Materials Science and Engineering at Zhejiang University. Her research focuses on HSMs, lithium-ion and lithium–sulfur batteries, all-solid-state batteries, and advanced ceramic matrix composites, developing energy materials for next-generation storage and conversion technologies.

Thomas Gennett is a Professor of Chemistry at Colorado School of Mines. He has extensive experience in the design, synthesis, characterization, and development of advanced HSM and systems development for hydrogen recovery.

Zaiping Guo is the Chair Professor of Energy Materials at City University of Hong Kong. Her research focuses on electrode materials, electrolytes, rechargeable batteries, hydrogen storage, fuel cells, and energy conversion, combining electrochemistry, materials science, and advanced characterization to develop next-generation sustainable energy technologies.

Ikutaro Hamada is a Professor at The University of Osaka. His work centers on the development of electronic structure methods and simulations of surfaces and interfaces, with applications ranging from fundamental phenomena to energy and environmental problems.

Jason Hattrick-Simpers is a Professor in the Department of Materials Science and Engineering, University of Toronto, and a Research Scientist at CanmetMATERIALS. He graduated with a B.S. in Mathematics and a B.S. in Physics from Rowan University and a PhD in Materials Science and Engineering from the University of Maryland.

Bjørn C. Hauback is a Chief Scientist at the Institute for Energy Technology (IFE) in Norway. His main competence is on complex and intermetallic hydrides, as well as understanding of the effect of catalysts in complex hydrides. He is an expert on neutron and X-ray diffraction methods.

Michael Hirscher is a Senior Scientist at the Max Planck Institute for Solid State Research. He has been active in the field of hydrogen storage in solids for more than 35 years. His particular focus is on MOFs for hydrogen storage and isotope separation.

Torben R. Jensen is a Professor of Chemistry and Doctor of Science (DSc) at Aarhus University. His research focuses on inorganic materials chemistry, HSMs, solid-state ionic conductors, and advanced energy materials. He combines synthesis and advanced materials characterization to develop functional materials for sustainable energy applications.

Baohua Jia, FTSE, is the Founding Director of RMIT's Centre for Atomaterials and Nanomanufacturing and Founding Editor-in-Chief of npj Nanophotonics. As an ARC Future Fellow and Optica Fellow, she has led A\$90 million in projects, published 400+ papers, and held 20+ patents.

Hyoungh Seop Kim is the Dean of the Graduate Institute of Ferrous & Eco Materials Technology (GIFT) and SeAh Chair Professor at POSTECH. His research focuses on physical metallurgy, high-entropy alloys, metal additive manufacturing, microstructure design, constitutive modeling, and AI-driven materials engineering for advanced structural materials.

Takahiro Kondo received his PhD in Engineering from the University of Tsukuba in 2003. After the postdoctoral research at RIKEN, he joined the University of Tsukuba in 2007. His research focuses on boron-based two-dimensional materials, particularly hydrogen boride and boron monosulfide nanosheets, for energy and environmental applications.

Kentaro Kutsukake is an Associate Professor at Nagoya University. He received his PhD from Tohoku University in 2007 and has held positions at JSPS, Tohoku University, Nagoya University, and RIKEN. His research focuses on applying data science to crystal engineering.

Xiao-Yan Li is an Assistant Professor in the Department of Chemistry at the National University of Singapore. Her group's research focuses on AI-integrated multiscale simulations for the efficient design and discovery of advanced functional nanomaterials.

Tongliang Liu is the Director of Sydney AI Centre at The University of Sydney. He is broadly interested in the fields of trustworthy ML, with emphases on learning with noisy labels, causal representation learning, and AI for science. He received the IEEE AI's 10 to Watch Award in 2022.

Piao Ma received his PhD from UT Austin and is currently a Scientist at the Gusu Laboratory of Materials and the CTO of Suzhou MatSource Technology Co. Ltd., a company dedicated to building a globally leading AI agent-driven high-throughput closed-loop platform for materials research and development.

Jianfeng Mao is an ARC Future Fellow and the Program Lead in Materials Engineering at the School of Chemical Engineering, Adelaide University. His current research interests are in developing advanced electrolytes and materials for next-generation batteries as well as the green battery recycling methods.

Rana Mohtadi is a Senior Principal Scientist at Toyota Research Institute of North America (TRINA) and a Principal Investigator at Tohoku University's Advanced Institute for Materials Research (WPI-AIMR), leading transformative materials research for chemical and electrochemical energy storage. She has published influential work and holds over 30 patents.

Hyunchul Oh is a Professor at UNIST, specializing in advanced nanoporous materials for hydrogen storage and isotope separation. He studied for his PhD at the Max Planck Institute in Germany. Previously, at Gyeongsang National University, his research focused on cryogenic gas storage and light isotope separation.

Mark Paskevicius is a Professor in Physics and Astronomy at Curtin University's Institute for Energy Transitions. His research centers on renewable energy storage, focusing on developing new materials for solid-state hydrogen storage. He also leads advanced research into next-generation solid-state batteries and the electrochemical production of hydrogen-rich materials.

Chris J. Pickard is the inaugural Sir Alan Cottrell Professor of Materials Science at Cambridge, using quantum mechanics to study matter at the atomic scale. As a lead CASTEP developer, he introduced the GIPAW method for NMR prediction and Ab Initio Random Structure Searching (AIRSS), and won the 2015 Rayleigh Medal.

Astrid Pundt is the Director at the Institute of Applied Materials (IAM), Karlsruhe Institute of Technology (KIT). She is head of the Metal and Materials Physics Division of the German Physical Society DPG and acts for the Deutsche Forschungsgemeinschaft DFG. She focusses on process–structure–properties relationships of defects (hydrogen) in materials.

Anibal Ramirez-Cuesta is the Neutron Instrument Development Group Leader at ORNL. PhD in Physics from the University of San Luis, Argentina. He is the author of the ICEMAN software and coauthor of the book *Vibrational Spectroscopy with Neutrons*. He is an expert in computational methods, data analysis, instrumentation, catalysis, hydrogen storage, porous materials, gas adsorption, etc.

Hiroyuki Saitoh received his PhD in Engineering and joined the Japan Atomic Energy Research Institute as a postdoctoral researcher in 2023. He is currently a Group Leader at QST. His research interests focus on metal hydrides and synchrotron radiation studies under high pressure.

Kaihang Shi is an Assistant Professor of Chemical and Biological Engineering at the University at Buffalo. His research integrates AI, molecular simulation, statistical mechanics, and mathematical modeling to discover, design, and characterize nanoporous materials and confined systems for energy and sustainability applications.

Aloysius Soon is a Full Professor in the Department of Materials Science and Engineering at Yonsei University and leads the Materials Theory Group. His research integrates first-principles calculations with ML to investigate surfaces, interfaces, and complex materials relevant to semiconductor and energy technologies.

Chenghua Sun is an Associate Professor of Chemistry at Swinburne University of Technology. His research focuses on computational materials chemistry, heterogeneous catalysis, electrocatalysis,

hydrogen technologies, carbon dioxide conversion, ammonia synthesis, and AI-driven catalyst design for clean energy and environmental applications.

Chris Wolverton is the Frank C. Engelhart Professor of Materials Science and Engineering at Northwestern University. His research focuses on computational materials science, first-principles modeling, materials informatics, ML, and high-throughput materials discovery, with applications to hydrogen storage, batteries, thermoelectrics, and sustainable energy materials.

Hiroshi Yabu received his PhD from Hokkaido University, Japan, in 2004. He started his own laboratory in Advanced Institute for Materials Research (WPI-AIMR), Tohoku University, as a Junior Principal Investigator in 2016. He is now a Full Professor and a Principal Investigator of WPI-AIMR, Tohoku University.

Weijie Yang received his PhD degree from North China Electric Power University in 2019. He is an Associate Professor at North China Electric Power University. His research focuses on mechanism- and data-driven HSM design and hydrogen energy storage systems.

Zhenpeng Yao is an Associate Professor at Shanghai Jiao Tong University whose research focuses on accelerating materials discovery through AI, autonomous experimentation, and computational materials science. His work integrates ML, robotics, and physics-based modeling to discover and optimize energy materials for sustainable energy storage and conversion.

Xuebin Yu is a Professor of Materials Science at Fudan University. His research focuses on the synthesis, characterization, and mechanistic understanding of lightweight HSMs, advancing materials for hydrogen energy applications through innovative design, experimental investigation, and materials chemistry.

Jianxin Zou is currently a Chair Professor at Shanghai Jiao Tong University. He is the Changjiang Chair Professor, a Fellow of the Royal Society of Chemistry, and a Fellow of the International Association of Advanced Materials. He is now mainly engaged in the preparation, characterization, and industrial applications of advanced magnesium-based energy materials.

Shouyi Hu earned his BS and MS in Chemical Engineering from Shanghai Jiao Tong University. Now a PhD student supervised by Prof. Zhigang Hu and Prof. Jianxin Zou, he studies ML screening of HSMs and system simulation.

Panpan Zhou received his PhD degree in Materials Science from Zhejiang University, China, in 2024. His research interests include medium- and high-entropy HSM customization, hydrogen evolution catalysts synthesis, first-principles calculation, and AI.

Xi Lin is a Research Assistant Professor at Shanghai Jiao Tong University. He is mainly engaged in solid-state HSMs, system simulation, and development.

Zhigang Hu, Associate Professor at Shanghai Jiao Tong University, previously researched at Cambridge and Brunel University. His research covers porous materials and catalysts for hydrogen/ammonia energy. He has published 85+ high-level papers, led more than 10 national projects, received major awards, and served as a deputy editor and associate editor.

Zhenhao Zhou is a PhD candidate in Materials Science and Engineering at Shanghai Jiao Tong University. His research focuses on interpretable ML for materials discovery, particularly nanoporous materials and crystalline defects.

Pengfei Ou is an Assistant Professor and Presidential Young Professor in the Department of Chemistry at the National University of Singapore. His research focuses on computational catalysis and AI for accelerating catalyst discovery and mechanistic understanding in energy and environmental transformations.

Jiayu Peng is an Assistant Professor of Materials Design and Innovation at the University at Buffalo, The State University of New York. He obtained his PhD in Materials Science and Engineering from MIT in 2022. His group uses AI to understand and design materials for chemical transformation and energy technologies.

Shin-ichi Orimo is the Director of the Advanced Institute for Materials Research (WPI-AIMR) and a Professor at the Institute for Materials Research at Tohoku University. He approaches hydrogen science from multifaceted perspectives, including high-density hydrogen storage, superionic conduction, superconductivity, data science, and robotics.

Hao Li is a Distinguished Professor at the Advanced Institute for Materials Research (WPI-AIMR), Tohoku University, Japan. His research focuses on developing AI, materials theory, and autonomous experimentation for closed-loop materials design. He proposed the concept of the digital materials ecosystem and serves as the founding Editor-in-Chief of the journal *AI Agent*. Webpage: <https://www.li-lab-cat-design.com/>.

■ ACKNOWLEDGMENTS

This work was supported by the Green Technologies of Excellence (GteX) Program, Japan (Grant No. JPMJGX23H1). M.B. acknowledges support from the Project CH4.0 under the MUR program “Dipartimenti di Eccellenza 2023-2027” (CUP: D13C22003520001). H.O. acknowledges that this work was partially supported by the National Research Foundation of Korea (NRF) grant funded by the Korea government (MSIT and MOE) (RS-2025-16063688). The authors acknowledge the important contribution of the International Energy Agency (IEA) Hydrogen Technology Collaboration Programme (TCP) Task 51: Hydrogen Materials for Energy. We thank Mr. Cheng Fang for his valuable assistance during the preparation of this manuscript.

■ REFERENCES

- (1) Mohtadi, R.; Orimo, S. i. The renaissance of hydrides as energy materials. *Nat. Rev. Mater.* **2017**, *2*, 16091.
- (2) Broom, D. P. et al. Hydrogen-based materials for energy storage and conversion. *Nat. Rev. Clean Technol.* **2026**, *2*, 512–533.
- (3) Webb, C. J.; Humphries, T. D.; Teprovich, J. A.; Polanski, M.; Heere, M.; Li, H. W.; Felderhoff, M.; Filinchuk, Y.; Dornheim, M.; Paskevicius, M.; Buckley, C. E.; Jensen, T. R. Diverse hydrogen chemistry with perspectives for energy storage. *Chem. Commun.* **2026**, *62*, 4477–4495.
- (4) Furukawa, H.; Cordova, K. E.; O’Keeffe, M.; Yaghi, O. M. The chemistry and applications of metal-organic frameworks. *Science* **2013**, *341*, 1230444.
- (5) Chen, Z.; Kirlikovali, K. O.; Idrees, K. B.; Wasson, M. C.; Farha, O. K. Porous materials for hydrogen storage. *Chem* **2022**, *8*, 693–716.
- (6) Sutton, A. L.; Mardel, J. I.; Hill, M. R. Metal-organic frameworks (MOFs) as hydrogen storage materials at near-ambient temperature. *Chem. –Eur. J.* **2024**, *30*, e202400717.
- (7) Zhang, R.; Daglar, H.; Tang, C.; Li, P.; Feng, L.; Han, H.; Wu, G.; Limketkai, B. N.; Wu, Y.; Yang, S.; Chen, A. X. Y.; Stern, C. L.; Malliakas, C. D.; Snurr, R. Q.; Stoddart, J. F. et al. Balancing volumetric and gravimetric capacity for hydrogen in supramolecular crystals. *Nat. Chem.* **2024**, *16*, 1982–1988.
- (8) Züttel, A. Materials for hydrogen storage. *Mater. Today* **2003**, *6*, 24–33.
- (9) Orimo, S. i.; Nakamori, Y.; Eliseo, J. R.; Züttel, A.; Jensen, C. M. Complex hydrides for hydrogen storage. *Chem. Rev.* **2007**, *107*, 4111–4132.
- (10) Hirscher, M. et al. Materials for hydrogen-based energy storage – Past, recent progress and future outlook. *J. Alloys Compd.* **2020**, *827*, 153548.
- (11) Jiang, M.; Song, A.; Yan, Y.; Feng, W.; Li, H.; Liang, B. Applicability and limitations of hydrogen affinity as a descriptor for designing high-entropy alloys for hydrogen storage. *Chem. Mater.* **2026**, *38*, 3666–3675.
- (12) Gebretatios, A. G.; Banat, F.; Cheng, C. K. A critical review of hydrogen storage: Toward the nanoconfinement of complex hydrides from the synthesis and characterization perspectives. *Sustain. Energy Fuels* **2024**, *8*, 5091–5130.
- (13) Witman, M. D.; Brooks, K. P.; Sprik, S. J.; Wood, B. C.; Heo, T. W.; Ray, K. G.; Klebanoff, L. E.; Acosta, A.; Stavila, V.; Allendorf, M. D. A bulk versus nanoscale hydrogen storage paradox revealed by material-system co-design. *Adv. Funct. Mater.* **2024**, *34*, 2411763.
- (14) Wang, Q.; Sato, R.; García-Méndez, R.; Jang, W.; Ou, P.; Soon, A.; Zhao, J.; Wang, X.; Orimo, S. i.; Cheng, E. J. AI agents for solid electrolytes: Opportunities, challenges, and future directions. *AI Agent* **2025**, *1*, 10.
- (15) Zhao, C.; Li, H. AI agents: Opportunity, hype, and the way through. *AI Agent* **2026**, *2*, 3.
- (16) Zhang, D.; Jia, X.; Wang, Y.; Liu, H.; Wang, Q.; Jang, S. H.; Shah, D.; Ye, S.; Tran, H. B.; Li, H. Digital materials ecosystem: from databases to AI agents for autonomous discovery. *Chem. Sci.* **2026**, *17*, 5782–5804.
- (17) Wang, X.; Li, Z.; Zhang, D.; Li, H.; Xu, H.; Cheng, D. Catalysis AI agent guides discovering the universal design principle of Cu-based single-atom alloy catalysts for CO₂ electroreduction. *Angew. Chem., Int. Ed.* **2026**, *65*, e24612.
- (18) Zhang, D.; Chen, Y.; Liu, C.; Liu, Y.; Xin, H.; Peng, J.; Ou, P.; Li, H. Accelerating catalyst materials discovery with large artificial intelligence models. *Angew. Chem., Int. Ed.* **2026**, *65*, e26150.
- (19) Gao, Z.; Yang, X.; Zhuang, Z.; Zhang, Y.; Cai, J.; Li, Y.; Fu, W.; Li, H.; Yang, W. Catalytic strategies and mechanisms for enhancing MgH₂ solid-state hydrogen storage. *Chem. Catal.* **2026**, *6*, 101692.
- (20) Cheng, M.; Fu, C. L.; Okabe, R.; Chottrattanapituk, A.; Boonkird, A.; Hung, N. T.; Li, M. Artificial intelligence-driven approaches for materials design and discovery. *Nat. Mater.* **2026**, *25*, 174–190.
- (21) Curtarolo, S.; Setyawan, W.; Hart, G. L. W.; Jahnatek, M.; Chepulskii, R. V.; Taylor, R. H.; Wang, S.; Xue, J.; Yang, K.; Levy, O.; Mehl, M. J.; Stokes, H. T.; Demchenko, D. O.; Morgan, D. AFLOW: An automatic framework for high-throughput materials discovery. *Comput. Mater. Sci.* **2012**, *58*, 218–226.
- (22) Saal, J. E.; Kirklin, S.; Aykol, M.; Meredig, B.; Wolverton, C. Materials design and discovery with high-throughput density functional theory: The open quantum materials database (OQMD). *JOM* **2013**, *65*, 1501–1509.
- (23) Sbailò, L.; Fekete, Á.; Ghiringhelli, L. M.; Scheffler, M. The NOMAD Artificial-Intelligence Toolkit: turning materials-science data into knowledge and understanding. *npj Comput. Mater.* **2022**, *8*, 250.
- (24) Horton, M. K.; Huck, P.; Yang, R. X.; Munro, J. M.; Dwaraknath, S.; Ganose, A. M.; Kingsbury, R. S.; Wen, M.; Shen, J. X.; Mathis, T. S.; Kaplan, A. D.; Berket, K.; Riebesell, J.; George, J.; Rosen, A. S.; Spotte-Smith, E. W. C.; McDermott, M. J.; Cohen, O. A.; Dunn, A.; Kuner, M. C.; Rignanese, G. M.; Petretto, G.; Waroquiers, D.; Griffin, S. M.; Neaton, J. B.; Chrzan, D. C.; Asta, M.; Hautier, G.; Cholia, S.; Ceder, G.; Ong, S. P.; Jain, A.; Persson, K. A. Accelerated data-driven materials science with the Materials Project. *Nat. Mater.* **2025**, *24*, 1522–1532.
- (25) Cavnignac, T.; Schmidt, J.; De Breuck, P. P.; Loew, A.; Cerqueira, T. F. T.; Wang, H. C.; Bochkarev, A.; Lysogorskiy, Y.; Romero, A. H.; Drautz, R.; Botti, S.; Marques, M. A. L. AI-Driven expansion and application of the Alexandria database. *J. Phys. Mater.* **2026**, *9*, 025014.
- (26) Chung, Y. G.; Haldoupis, E.; Bucior, B. J.; Haranczyk, M.; Lee, S.; Zhang, H.; Vogiatzis, K. D.; Milisavljevic, M.; Ling, S.; Camp, J. S.; Slater, B.; Siepmann, J. I.; Sholl, D. S.; Snurr, R. Q. Advances,

Updates, and analytics for the computation-ready, experimental metal-organic framework database: CoRE MOF 2019. *J. Chem. Eng. Data* **2019**, *64*, 5985–5998.

(27) Rosen, A. S.; Fung, V.; Huck, P.; O'Donnell, C. T.; Horton, M. K.; Truhlar, D. G.; Persson, K. A.; Notestein, J. M.; Snurr, R. Q.; et al. High-throughput predictions of metal-organic framework electronic properties: Theoretical challenges, graph neural networks, and data exploration. *npj Comput. Mater.* **2022**, *8*, 112.

(28) Bobbitt, N. S.; Shi, K.; Bucior, B. J.; Chen, H.; Tracy-Amoroso, N.; Li, Z.; Sun, Y.; Merlin, J. H.; Siepmann, J. I.; Siderius, D. W.; Snurr, R. Q. MOFDB: An Online Database of Computational Adsorption Data for Nanoporous Materials. *J. Chem. Eng. Data* **2023**, *68*, 483–498.

(29) Allendorf, M. D.; Hulvey, Z.; Gennett, T.; Ahmed, A.; Autrey, T.; Camp, J.; Seon Cho, E.; Furukawa, H.; Haranczyk, M.; Head-Gordon, M.; Jeong, S.; Karkamkar, A.; Liu, D. J.; Long, J. R.; Meihaus, K. R.; Nayyar, I. H.; Nazarov, R.; Siegel, D. J.; Stavila, V.; Urban, J. J.; Veccham, S. P.; Wood, B. C. An assessment of strategies for the development of solid-state adsorbents for vehicular hydrogen storage. *Energy Environ. Sci.* **2018**, *11*, 2784–2812.

(30) Mendoza-Cortés, J. L.; Han, S. S.; Goddard, W. A. High H₂ uptake in Li-, Na-, K-metalated covalent organic frameworks and metal organic frameworks at 298 K. *J. Phys. Chem. A* **2012**, *116*, 1621–1631.

(31) Kim, W. T.; Lee, W. G.; An, H. E.; Furukawa, H.; Jeong, W.; Kim, S. C.; Long, J. R.; Jeong, S.; Lee, J. H. Machine learning-assisted design of metal-organic frameworks for hydrogen storage: A high-throughput screening and experimental approach. *Chem. Eng. J.* **2025**, *S07*, 160766.

(32) Livas, C. G.; Trikalitis, P. N.; Froudakis, G. E. MOFSynth: A computational tool toward synthetic likelihood predictions of MOFs. *J. Chem. Inf. Model.* **2024**, *64*, 8193–8200.

(33) Sarikas, A. P.; Gkagkas, K.; Froudakis, G. E. RetNeXt: A pre-trained model for transfer learning across the MOF adsorption space. *J. Chem. Inf. Model.* **2026**, *66*, 2110–2116.

(34) Jang, S.-H.; Zhang, D.; Jia, X.; Tran, B. H.; Zhang, L.; Sato, R.; Hashimoto, Y.; Sato, T.; Konno, K.; Orimo, S.; Li, H. Digital Hydrogen Platform (DigHyd): A rigorously curated database for hydrogen storage materials empowered by AI-assisted literature mining. *arXiv:2603.14139*, **2026**, 10.48550/arXiv.2603.14139.

(35) Zhang, D.; Jia, X.; Tran, H. B.; Jang, S. H.; Zhang, L.; Sato, R.; Hashimoto, Y.; Sato, T.; Konno, K.; Orimo, S. i.; Li, H. “DIVE” into hydrogen storage materials discovery with AI agents. *Chem. Sci.* **2026**, *17*, 3031–3042.

(36) Evans, J. D.; Bon, V.; Senkovska, I.; Kaskel, S. A universal standard archive file for adsorption data. *Langmuir* **2021**, *37*, 4222–4226.

(37) Broom, D. P.; Hirscher, M. Improving reproducibility in hydrogen storage material research. *ChemPhysChem* **2021**, *22*, 2141–2157.

(38) Li, K.; DeCost, B.; Choudhary, K.; Greenwood, M.; Hattrick-Simpers, J. A critical examination of robustness and generalizability of machine learning prediction of materials properties. *npj Comput. Mater.* **2023**, *9*, 55.

(39) Jang, S. H.; Zhang, D.; Tran, H. B.; Jia, X.; Konno, K.; Sato, R.; Orimo, S. i.; Li, H. Physically interpretable descriptors drive the materials design of metal hydrides for hydrogen storage. *Chem. Sci.* **2025**, *16*, 23111–23120.

(40) Jang, S.-H. GoodRegressor: A hierarchical inductive bias for navigating high-dimensional compositional space. *arXiv:2510.18325*, **2026**, 10.48550/arXiv.2510.18325.

(41) Jang, S. H.; Zhang, D.; Jia, X.; Tran, H. B.; Zhang, L.; Sato, R.; Hashimoto, Y.; Ohashi, Y.; Sato, T.; Konno, K.; Orimo, S. i.; Li, H. A unified descriptor framework for hydrogen storage capacity and equilibrium pressure in interstitial hydrides. *Chem. Sci.* **2026**, *17*, 13396–13406.

(42) Li, C.; Yang, W.; Liu, H.; Liu, X.; Xing, X.; Gao, Z.; Dong, S.; Li, H. Picturing the gap between the performance and US-DOE's hydrogen storage target: A data-driven model for MgH₂ dehydrogenation. *Angew. Chem., Int. Ed.* **2024**, *63*, e202320151.

(43) Cai, J.; Yao, T.; Ma, P.; Fu, W.; Dong, S.; Zhang, D.; Zhang, L.; Gao, Z.; Li, H.; Yang, W. Dynamic kinetics of MgH₂ in solid-state

hydrogen storage: From the “dam-break effect” to mechanistic understanding, data-driven design, and application strategies. *Acc. Chem. Res.* **2026**, *59*, 2561–2574.

(44) Zhou, P.; Zhou, Q.; Xiao, X.; Fan, X.; Zou, Y.; Sun, L.; Jiang, J.; Song, D.; Chen, L. Machine learning in solid-state hydrogen storage materials: Challenges and perspectives. *Adv. Mater.* **2025**, *37*, 2413430.

(45) Klopčič, N.; Grimmer, I.; Winkler, F.; Sartory, M.; Trattner, A. A review on metal hydride materials for hydrogen storage. *J. Energy Storage* **2023**, *72*, 108456.

(46) Voskuilen, T. G.; Pourpoint, T. L. Phase field modeling of hydrogen transport and reaction in metal hydrides. *Int. J. Hydrogen Energy* **2013**, *38*, 7363–7375.

(47) Unke, O. T.; Chmiela, S.; Sauceda, H. E.; Gastegger, M.; Poltavsky, I.; Schütt, K. T.; Tkatchenko, A.; Müller, K. R. Machine learning force fields. *Chem. Rev.* **2021**, *121*, 10142–10186.

(48) Zhang, L.; Han, J.; Wang, H.; Car, R.; E, W. Deep potential molecular dynamics: A scalable model with the accuracy of quantum mechanics. *Phys. Rev. Lett.* **2018**, *120*, 143001.

(49) Cusentino, M. A.; Wood, M. A.; Thompson, A. P. Machine learned interatomic potentials for gas-metal interactions. *Model. Simul. Mater. Sci. Eng.* **2024**, *33*, 015007.

(50) Kulichenko, M.; Nebgen, B.; Lubbers, N.; Smith, J. S.; Barros, K.; Allen, A. E. A.; Habib, A.; Shinkle, E.; Fedik, N.; Li, Y. W.; Messerly, R. A.; Tretiak, S. Data Generation for Machine Learning Interatomic Potentials and Beyond. *Chem. Rev.* **2024**, *124*, 13681–13714.

(51) Yuan, E. C. Y.; Liu, Y.; Chen, J.; Zhong, P.; Raja, S.; Kreiman, T.; Vargas, S.; Xu, W.; Head-Gordon, M.; Yang, C.; Blau, S. M.; Cheng, B.; Krishnapriyan, A.; Head-Gordon, T. Foundation models for atomistic simulation of chemistry and materials. *Nat. Rev. Chem.* **2026**, *10*, 212–230.

(52) Angeletti, A.; Leoni, L.; Massa, D.; Pasquini, L.; Papanikolaou, S.; Franchini, C. Hydrogen diffusion in magnesium using machine learning potentials: a comparative study. *npj Comput. Mater.* **2025**, *11*, 85.

(53) Wang, N.; Huang, S. Molecular dynamics study on magnesium hydride nanoclusters with machine-learning interatomic potential. *Phys. Rev. B* **2020**, *102*, 094111.

(54) Ito, K.; Matsumura, N.; Iwasaki, Y.; Sakai, Y.; Yamamura, M.; Omura, T.; Yamabe, J.; Matsunaga, H. Predicting hydrogen diffusion in nickel-manganese random alloys using machine learning interatomic potentials. *Commun. Mater.* **2025**, *6*, 195.

(55) Shuang, F.; Ji, Y.; Wei, Z.; Dong, C.; Gao, W.; Laurenti, L.; Dey, P. Decoding the hidden dynamics of super-Arrhenius hydrogen diffusion in multi-principal element alloys via machine learning. *Acta Mater.* **2025**, *289*, 120924.

(56) Kumar, P.; Körmann, F.; Edalati, K.; Grabowski, B.; Ikeda, Y. Hydrogen diffusion in TiCr₂H_x Laves phases: A combined ab initio and machine-learning-potential study. *Acta Mater.* **2026**, *308*, 122048.

(57) Kumar, P.; Körmann, F.; Grabowski, B.; Ikeda, Y. Machine learning potentials for hydrogen absorption in TiCr₂ Laves phases. *Acta Mater.* **2025**, *297*, 121319.

(58) Qi, J.; Ko, T. W.; Wood, B. C.; Pham, T. A.; Ong, S. P. Robust training of machine learning interatomic potentials with dimensionality reduction and stratified sampling. *npj Comput. Mater.* **2024**, *10*, 43.

(59) Sato, R.; Conway, L. J.; Zhang, D.; Pickard, C. J.; Akagi, K.; Sau, K.; Li, H.; Orimo, S. i. Surface melting-driven hydrogen absorption for high-pressure polyhydride synthesis. *Proc. Natl. Acad. Sci. U. S. A.* **2025**, *122*, e2413480122.

(60) Zhang, B.; Asta, M.; Wang, L. W. Machine learning force field for Fe-H system and investigation on role of hydrogen on the crack propagation in α -Fe. *Comput. Mater. Sci.* **2022**, *214*, 111709.

(61) Ito, K.; Otaki, T.; Yokoi, T.; Hyodo, K.; Mori, H. Machine learning interatomic potential reveals hydrogen embrittlement origins at general grain boundaries in α -iron. *Commun. Mater.* **2026**, *7*, 30.

(62) Tahmasbi, H.; Ramakrishna, K.; Lokamani, M.; Cangi, A. Machine learning-driven structure prediction for iron hydrides. *Phys. Rev. Mater.* **2024**, *8*, 033803.

(63) Kwon, H.; Shiga, M.; Kimizuka, H.; Oda, T. Accurate description of hydrogen diffusivity in bcc metals using machine-learning moment

tensor potentials and path-integral methods. *Acta Mater.* **2023**, *247*, 118739.

(64) Kataoka, Y.; Haruyama, J.; Sugino, O.; Shiga, M. Predictive evaluation of hydrogen diffusion coefficient on Pd(111) surface by path integral simulations using neural network potential. *Phys. Rev. Res.* **2024**, *6*, 043224.

(65) Steffen, J.; Alibakhshi, A. Hydrogen diffusion on Ni(100): A combined machine-learning, ring polymer molecular dynamics, and kinetic Monte Carlo study. *J. Chem. Phys.* **2024**, *161*, 184116.

(66) Ba Tran, H.; Sato, T.; Sato, R.; Saitoh, H.; Orimo, S. I.; Li, H. Tuning stability of AB₃-type alloys by suppressing magnetism. *Chem. Mater.* **2026**, *38*, 497–507.

(67) Palumbo, M.; Dematteis, E. M.; Fenocchio, L.; Cacciamani, G.; Baricco, M. Advances in CALPHAD methodology for modeling hydrides: A comprehensive review. *J. Phase Equilib. Diffus.* **2024**, *45*, 273–289.

(68) Alvares, E.; Sellschopp, K.; Wang, B.; Kang, S.; Klassen, T.; Wood, B. C.; Heo, T. W.; Jerabek, P.; Pistidda, C. Multiscale modeling of metal-hydride interphases—quantification of decoupled chemo-mechanical energies. *npj Comput. Mater.* **2024**, *10*, 249.

(69) Heo, T. W.; Colas, K. B.; Motta, A. T.; Chen, L. Q. A phase-field model for hydride formation in polycrystalline metals: Application to δ -hydride in zirconium alloys. *Acta Mater.* **2019**, *181*, 262–277.

(70) Alekseeva, S.; Strach, M.; Nilsson, S.; Fritzsche, J.; Zhdanov, V. P.; Langhammer, C. Grain-growth mediated hydrogen sorption kinetics and compensation effect in single Pd nanoparticles. *Nat. Commun.* **2021**, *12*, 5427.

(71) Dyck, A.; Gisy, J.; Hille, F.; Wagner, S.; Pundt, A.; Böhlke, T. Hydride formation in open thin film metal hydrogen systems: Cahn-Hilliard-type phase-field simulations coupled to elasto-plastic deformations. *Mech. Mater.* **2025**, *203*, 105258.

(72) Zeni, C.; Pinsler, R.; Zügner, D.; Fowler, A.; Horton, M.; Fu, X.; Wang, Z.; Shysheya, A.; Crabbé, J.; Ueda, S.; Sordillo, R.; Sun, L.; Smith, J.; Nguyen, B.; Schulz, H.; Lewis, S.; Huang, C. W.; Lu, Z.; Zhou, Y.; Yang, H.; Hao, H.; Li, J.; Yang, C.; Li, W.; Tomioka, R.; Xie, T. A generative model for inorganic materials design. *Nature* **2025**, *639*, 624–632.

(73) Wilmer, C. E.; Leaf, M.; Lee, C. Y.; Farha, O. K.; Hauser, B. G.; Hupp, J. T.; Snurr, R. Q. Large-scale screening of hypothetical metal-organic frameworks. *Nat. Chem.* **2012**, *4*, 83–89.

(74) Cissé, A.; Cooper, M. E.; Zhu, M.; Evangelopoulos, X.; Cooper, A. I. Can we automate scientific reasoning in closed-loop experiments using large language models? *Digit. Discov.* **2026**, *5*, 1132–1160.

(75) Wang, H.; Castañeda, R. E.; Werber, J. R.; Fehlis, Y.; Kim, E.; Hatrick-Simpers, J. Training-free active learning framework in materials science with large language models. *npj Comput. Mater.* **2026**.

(76) Campos dos Santos, E.; Sato, R.; Kisu, K.; Sau, K.; Jia, X.; Yang, F.; Orimo, S. i.; Li, H. Explore the ionic conductivity trends on B₁₂H₁₂ divalent closo-type complex hydride electrolytes. *Chem. Mater.* **2023**, *35*, 5996–6005.

(77) Wang, Q.; Yang, F.; Wang, Y.; Zhang, D.; Sato, R.; Zhang, L.; Cheng, E. J.; Yan, Y.; Chen, Y.; Kisu, K.; Orimo, S.; Li, H. Unraveling the complexity of divalent hydride electrolytes in solid-state batteries via a data-driven framework with large language model. *Angew. Chem., Int. Ed.* **2025**, *64*, e202506573.

(78) Zou, Y.; Cheng, A. H.; Aldossary, A.; Bai, J.; Leong, S. X.; Campos-Gonzalez-Angulo, J. A.; Choi, C.; Ser, C. T.; Tom, G.; Wang, A.; Zhang, Z.; Yakavets, I.; Hao, H.; Crebolder, C.; Bernalles, V.; Aspuru-Guzik, A. El Agente: An autonomous agent for quantum chemistry. *Matter* **2025**, *8*, 102263.

(79) Kusne, A. G.; Yu, H.; Wu, C.; Zhang, H.; Hatrick-Simpers, J.; DeCost, B.; Sarker, S.; Oses, C.; Toher, C.; Curtarolo, S.; Davydov, A. V.; Agarwal, R.; Bendersky, L. A.; Li, M.; Mehta, A.; Takeuchi, I. On-the-fly closed-loop materials discovery via Bayesian active learning. *Nat. Commun.* **2020**, *11*, 5966.

(80) Bennett, J. A.; Orouji, N.; Khan, M.; Sadeghi, S.; Rodgers, J.; Abolhasani, M. Autonomous reaction Pareto-front mapping with a self-driving catalysis laboratory. *Nat. Chem. Eng.* **2024**, *1*, 240–250.

(81) Cao, Z.; Wang, L. Reinforcement fine-tuning for materials design. *Phys. Rev. B* **2026**, *113*, 024106.